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MATERIALS REQUIREMENTS FOR ADVANCED ENERGY SYSTEMS - NEW FUELS. VOLUME 3: MATERIALS RESEARCH NEEDS IN ADVANCED ENERGY SYSTEMS USING NEW FUELS

N. H. G. Daniels, et al

Stanford Research Institute

Prepared for:

Defense Supply Service Advanced Research Projects Agency

July 1974

DISTRIBUTED BY:



5285 Port Royal Road, Springfield Va. 22151

SECURITY CLASSIFICATION OF THIS PAGE (When Data Entered)

REPORT DOCUMENTATIO	BEFORE COMPLE					
1. REPORT NUMBER	2. GOV? ACCESSION NO	3. RECIPIENT'S CATAL	OG NUMBER			
		11)/A- 00	4550			
4. TITLE (and Subtitle)		5. TYPE OF REPORT &	PERIOD COVERED			
Materials Requirements for Advanced	Final Report					
New Fuels		1 May 1972 - 3	3). July 1974			
Vol. 3: Materials Research Need		6. PERFORMING ORG.	DEBOOT NUMBER			
Energy Systems Using No.	ew Fuels	PYU 2580	REPORT NUMBER			
N.H.G. Daniels, B. C. Syrett, and	l D T Towns	B. CONTRACT OR GRA	NT NUMBER(s)			
win.d. banters, b. c. Syrect, and	r. b. Jones	DAHC 15 73 C C	313			
9. PERFORMING ORGANIZATION NAME AND ADDI	RESS	10. PROGRAM ELEMEN AREA & WORK UNI	T, PROJECT, TASK			
Stanford Research Institute		ARPA Order No.				
Menlo Park, California 94025			1			
		12. REPORT DATE	ode No. A74880			
11. CONTROLLING OFFICE NAME AND ADDRESS Defense Supply Service-Washington		July 1974	114			
-	<u>l</u>	15. SECURITY CLASS. (of this report)			
Room 1D 245, The Pentagon		Unclassified				
Washington, D.C. 20310 14 MONITORING AGENCY NAME & ADDRESS (if d	iff from Controlling Office)					
Advanced Research Projects Agency	=					
1400 Wilson Boulevard, Arlington,		15a. DECLASSIFICATION SCHEDULE	N/DOWNGRADING			
	111611111 22000					
16. DISTHIBUTION STATEMENT (of this report)						
		DISTRIBUTION S	PATEMENT A			
		Approved for pu	blic releans;			
		Distribution 1	- 1			
		Company of the same of the sam				
17. DISTRIBUTION STATEMENT (of the abstract ente	red in Biock 20, if different f	rom report)				
18. SUPPLEMENTARY NOTES						
•						
19. KEY WORDS (Continue on reverse side if necessary	and identify by block number	r)				
Energy	Research					
Nonfossil Fuels	Planning					
Materials						
Interactions						
20. ABSTRACT (Continue on reverse side if necessary a	nd identify by block number)					
This program sought to identi	fy materials-critic	cal aspects of the	e use, produc-			
tion, transportation, and storage	of new fuels derive	ed from nonfossil	sources.			
Hydrogen was the principal new fuel studied; hydrogen-derived fuels considered were						
ammonia, hydrazine, boranes, silanes, carbon monoxide, and methyl alcohol. The						
materials implications of the use		•				
generation) as a fuel oxidizer and						
also examined during the program.						

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19. KEY WORDS (Continued)

20 ABSTRACT (Continued)

The report consists of three volumes

Volume 1: Interactions of Materials and New Fuels
Volume 2: Materials Aspects of the Use, Production,
Transportation and Storage of New Yuels

Volume 3: Materials Research Needs in Advanced Energy Systems Using New Fuels.

Materials research programs are recommended in the areas of hydrogen/materials interactions, high-temperature materials, properties of materials at cryogenic temperatures, materials for electrochemical systems (fuel cells, batteries, electrolysers), and catalysis. It is concluded that while a significant materials research and testing effort is needed in support of advanced energy systems using new fuels, materials requirements will not constitute a major impediment to the implementation of such systems.

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FOREWORD

This report contains the results of a research study supported by the Advanced Research Projects Agency under ARPA order No. 2484. The ARPA Project Officer was Dr. Stanley Ruby, Materials Sciences.

This report is presented in three volumes, containing the following major sections.

- Volume 1. Interactions of Materials with New Fuels
 - I Introduction
 - II General Characteristics of New Fuels
 - III Behavior of Engineering Materials in New Fuel Environments
- Volume 2. Materials Aspects of the Use, Production, Transportation and Storage of New Fuels
 - IV Materials Aspects of the Use of New Fuels
 - V Materials Aspects of the Production of New Fuels from Nonfossil Sources
 - VI Materials Aspects of the Transportation of New Fuels
 - VII Materials Aspects of the Storage of New Fuels
- Volume 3. Materials Research Needs in Advanced Energy Systems
 Using New Fuels
 - VIII Correlation and Analysis of Materials Requirements
 - IX Research Recommendations and Priorities

The authors of this report would like to acknowledge the valuable contributions to the performance of this study by the following SRI staff members: T. Anyos, M. Barnes, E. Capener, T. Goodale, D. Hildenbrand, P. Jorgensen, G. Koo, R. Weaver, H. Wise, R. Wright, and the Staff of the Report Services Department. We would especially like to acknowledge the special assistance of Dr. J. Giner, Consultant, with

regard to the discussions of fuel cells, the electrolytic production of hydrogen, and high energy density batteries. We would also like to express our appreciation of the many members of the staff of government and industrial organizations who gave us the benefit of their knowledge and experience.

VIII CORRELATION AND ANALYSIS OF MATERIALS REQUIREMENTS

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VIII CORRELATION AND ANALYSIS OF MATERIALS REQUIREMENTS

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VIII CORRELATION AND ANALYSIS OF MATERIALS REQUIREMENTS

This section summarizes the materials research, development and testing programs identified in Sections IV through VII (Volume 2) needed to support the use, production, transportation, and storage of the new fuels. The summary is presented in tabular form in Tables VIII-1 through VIII-4. Table VIII-1 summarizes the materials research, development, and testing needs related to the use of new fuels and Table VIII-2 provides this information for the production of new fuels. Tables VIII-3 and VIII-4 summarize the transportation and storage aspects, respectively, of the study.

The primary grouping used in these tables is the major equipment or process class. The significance of the various column headings is described in more detail below.

Item Number (Column 1)

The item number appearing in the first column is a three-digit number that identifies a specific research, development, or testing need listed in Column 7. The first of the three digits identifies the major equipment or process class to which the item relates. For example, in Table VIII-1 the initial digit 1 refers to the major equipment class of Turbines. The second digit is allocated to a subclass of equipment or process associated with a particular fuel. For example, the second digit 1 in Table VIII-1 is associated with conventional steam turbines burning hydrogen, while the second digit 2 is associated with conventional gas turbines burning hydrogen with air; the second digit 5 is associated with conventional gas turbines burning ammonic. The final digit refers to a specific materials research, development, or testing need itemized in Column 7 of the table.

Equipment Class (Column 2)

The equipment or process class is shown in Column 2 of the table, with the actual heading varying according to the major topic--use-production, transportation, or storage--with which the table is concerned. The major equipment or process class is shown capitalized in association with a single digit in Column 1. Under each major class a number of subclasses of equipment or processes are listed to further define the problems and solutions associated with each major class. For example, in Table VIII-2 under the major process class of Advanced Electrolyzers, the various electrolyzer types are subclassified according to the electrolyte used. Each electrolyte is indicated by a different second digit in the item number, as explained above.

Fuel (Column 3)

Column 3 lists the particular fuel with which the research, development, and testing need is associated. In some instances, where there is an option of burning the fuel with air or oxygen, the use of one or the other is noted.

Problem Area (Column 4)

Column 4 shows the general problem area in which the research, development, and testing need exists. This is generally a particular part of the specific equipment or process listed in Column 2. In some instances, it is a generalized statement of a problem, which may or may not have particular materials aspects. For example, in Table VIII-2 conventional electrolyzers of both unipolar (Item 1.1.1) and bipolar (Item 1.2.1) types do not suffer from any major materials problem but are generally deficient in the problem area of efficiency and cost, for which the only solution appears to be a major improvement in the overall technology.

Type of Solution (Column 5)

Column 5 defines the type of activity needed to solve the problem described in Column 4. This may be a materials activity described by general terms, such as materials research or materials selection, or by more specific terms, such as catalyst development. In some instances the activity required may be of a type, such as engineering design, to which materials oriented activities do not directly contribute. In many instances, the type of solution involves both materials oriented and other types of activity.

Materials Problems (Column 6)

Column 6 identifies specific materials problems, that are components of the general problem area listed in Column 4. Several materials problems may, of course, be identified within a single problem area. In other instances, such as those where the type of solution indicated in Column 5 does not include a materials activity, Column 6 contains a negative entry, such as "none" or "none expected."

Materials R, D, and T Needs (Column 7)

Column 7 lists the specific program required to solve the materials problems identified in Column 6. Where possible these programs are described in sufficient detail to identify the specific materials activity. In some instances, an entry is made that indicates general support for engineering design or development activities. In these cases, the precise nature of the materials research, development, and testing needs cannot be identified until the engineering effort is in progress. Each entry in this column is specifically identified by the item number in Column 1, even when no specific materials research, development, and testing needs exist.

Remarks (Column 8)

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Column 8 contains comments to clarify the problem area or the program suggested. In addition, references are made to other items in which similar or related work is suggested. No comment is indicated by a dash.

Report References (Column 9)

Column 9 is a specific topic reference to Volume 2 of the report.

Additional information and discussion of each problem area can be located in Volume 2 by using this topic reference in Conjunction with the Table of Contents of the relevant report section.

TABLE VIII-1 MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF NEW FUELS

Table VI.I-1

MATLICIALS RESEARCH DEVELOPMENT AND TESTING MEEDED TO SUPPORT THE USE

Item No.	Equipment Class	Fue1	Problem Area	Type of Solution	Materials Problems	Material
1.	TURBINES					
1.1.1	Steam, conventional.	N ₂ 'air.	Fuel supply system exposed to hydrogen.	Materials selection.	${\rm H_2}$ environment effects.	Test materi environment operating l
1.1.2			Modified burners required. Righer flowe tempera- ture.	Pea(gn: Satelials selection.	Increased operating	Component t
1.2.1	Gas, conventional (including industrial, aircraft, marine, and automotive types).	H ₂ (gas or liquid) /air.	Fuel supply systems, heat exchanger/garifier exposed to hydrogen.	Materials selection,	Cryogenic temperatures, ${\rm H_2}$ environment effects.	Determine t erties of c purity and H ₂ gas envi approximate
1.2.2			Modifications required for compressor/turbine matching.	Design.	None expected.	None.
1.2.3			Combustors, vanes, bludes, affected by higher clame temperatures, high H ₂ O content combustion gases.	Materials research, development and testing.	Effects of high-temperature, high H_2O content environments on combustor and turbine materials.	Fundamental temperature on Si ₃ N ₄ , C
1.2.4						Determine c and oxidati candidate a ings in sim gases.
1.2.5				Sagineering design and development of advanced cooling sethods.	Materials behavior in fuel or water-cooled combustors, vancs or blades.	Erosion/cor and ceramic temperature
1.2.6						Behavior of ceramics in mately 2500
1.2.7					Fabrication of components for advanced cooling systems.	Materials of design and cooling sys
11311	Gar, hydrogen expansion.	H ₂ liquid, gasified and burned with air.	Fuel supply system, heat exchanger/gasi- fler, and turbine exposed to hydrogen.	Naterials selection.	Effects of high purity H ₂ environment on materials of construction from -423°F to 1500°F under steady and fluctuating stresses.	Materials a long times Effect of H ture toughn of stainles

,这是这些是是是是是是是是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种,我们就是这种的, 第一

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Table VIII-1 **ARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF NEW FUELS

& Solution	Materials Problems	Materials R, D, and T Needs	Fomerka	Report Reference in Vol. 2, Sect. IV
selection.	H ₂ environment effects.	Test materials and components in H_2 environments to establish safe operating limits.		A-1.1, A-4.4, A-5.4
muterials	Increased operating temperature.	Component testirt.	Life of boiler tubing likely to increase due to clean fuel.	A-1
A selection.	Cryogenic temperatures, ${\rm H_2}$ environment effects.	Determine tensile and fatigue properties of candidate materials in nigh purity and deliberately contaminated H ₂ gas environments from -423°F to approximately 600°F.	Candidate materials include Al alloys austenitic stain- less ricels, nickel alloys, and special brazing alloys,	A-2.2, A-4.4, A-5.4 (See also Vol. 1, Section III-A).
જો કું	None expected.	None.	Advanced composite and Ti alloy materials and fabric- ation programs for conven- tionally fueled gas turbines are relevant.	A-2.1, A-4.1
selection. suterials selection. research, and	Effects of high-temperature, high H ₂ O content environments on combustor ar urbine materials.	Fundamental studies of effect of high-temperature , high- $\rm H_2O$ environments on $\rm Si_3N_4$, Cb ulloys, coatings.	Development and testing of high-temperature alloys, coatings and ceramics, pro- ceeding for conventionally fueled gas turbines, is relevant. R, D, and T pro- grams need expanding to include modified environ- ments due to change of fuel.	A-2, A-4, A-5.2
e de la companya de l		Determine creep and fatigue properties and oxidation/corrosion resistance of candidate alloys, ceramics and coatings in simulated H ₂ /air combustion gases.		A-2, A-4, A-5.2
ing design and ent of advanced methods.	Materials behavior in fuel or water-cooled combustors, vanes or blades.	Erosion/corrosion of candidate alloys and ceramics in high-velocity, high-temperature $\hat{h_3}$ 0.	Required if water cooling is feasible.	A-2, A-4, A-5.2
A STATE OF THE STA		Behavior of candidate alloys and ceramics in H ₂ from 123° to approximutely 2590°F.	Required if hydrogen fuel cooling is feasible.	A-2, A-4, A-5.2
	Fabrication of components for advanced cooling systems.	Materials engineering sumport for design and development of advanced cooling systems.	_	A-2,A-4, A-5.2
Selection.	Effects of high purity H ₃ convironment on materials of construction from ~123°F to 1500°F under steady and fluctuating stresses.	Materials and component testing for long times in high-purity H ₂ gas. Effect of H ₂ on tansile, crosp, fracture toughness and fatigue behavior of stainless steels, superalloys and brazing alleys from -425°F to 1500°F.	Overlaps items 1.1.1 and 1.3.1.	A-2.2. A-4.3.5, A-4.4, A-5.4
Selection.	VIII-5			

Table VIII-1 (Coatinued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING MEEDED TO SUPPORT THE USE OF

Item	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problems	Maioriuls I
1.	TURBINES (Con't)					
1.4.1	H-drogen/oxygen.	H ₂ (gas or liquid)/O ₂ (gas or liquid).	H ₂ supply system.	Materials selection.	Cryogonic temperatures, H ₂ environment effects.	See item 1.2.
1.4.2			0_2 supply system.	Establish and adhere to safe y standards.	Ignition of metals and organic materials.	None.
1.4.3		~~~~	Combustors and turbine components exposed to very high temperatures and high-temperature H ₂ C environment.	Engineering design and development of advanced cooling methods. Materials R, D, and T.	Effects of high temperature and H ₂ O environments on candidate metallic and seramic materials of conconstruction.	None at this t
1.5.1	Gas, conventional.	NH ₃	Fuel supply system.	Materials selection.	Stress corrosion cracking of steels, copper alloys.	Establish limingard o steel evel and type nant concentrations.
1.5.2			Compressor and turbine.	Design for optimum performance.	No special problems expected.	None.
1.6.1	Gas, conventional.	Methanol.	Fuel supply system.	Materials selection.	Avoid Ti alloys.	None.
1,6.2			Compressor and turbine.	Design for optimum performance.	None.	None.
1.7,1	Gas, conventional.	co	No problems expected in fuel supply Systems.	-	None.	None.
1.7.2			Compressor and turbine.	Design for optimum performance.	None.	None.
2.	HYVERSONIC AIRCRAFT ENGINES				/	
2.1.1	Scramjet.	H ₂ /sir	Engines and heat exchangers.	Design, materials selection, development and testing.	High-ressure hydrogen environments from cryogenic to high temperatures; high- terpe-ature oxidizing environments, high aerody- namic stresses, high-fre- quency faligue, thermal fatigue, severe thermal shock and thermal stressos.	Not clearly d

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RESEARCH DEVELOPME	NT AND TECTING RESENTED TO SUPP	ORT THE USF (F NEW FUELS		
X				
Dens of Polytics	Manager and a Dunkland	Manager of the the and the Manager	Rean*ks	Report Reference
Type of Solution	Materials Problems	Materiais R. D. and T Necas	ROBATAS	in Vol. 2, Eect. 1
erials selection.	Cry genic temperatures,	See item 1.2.1.	See ites 1.º.1.	A-1.3, A-5.3
	H, environment effects.			,
	-			
teblish and adhere	Ignition of metals and	None.	-	See Vol. 1, Sect.
safety standards.	organic materials.			III-F
incering design and	Effects of high temperature	None at this time.	Initiation of materials R,	A-1.3, A-4.2, A-5
	and H ₂ O environments on		b, and T should await further	
ling methods.	candidate metallin and		concept development. Items	
teriule R, D, and T.	ceramic materials of con-		1.2.3. through 1.2.7 will	
	construction.		generate relevant data.	
				1 0 O
erials selection.	Stress corrosion cracking of steels, copper alloys.	Establish limits of phenomena with regard to stool composition, stress	Research programs primarily directed to materials prob-	A-3. See also Vol. 1, Sect. III
	or steers, topper arroys.	level and type, and ammonia contami-	leus in the transportation of	
		nant concentration.	ammonia will provide most of	
			this data.	
sign for optimus	No special problems expec-	None.	-	A-3
formance.	ted.			
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terials selection.	Ávoid Ti ailoys.	None.	-	A-3. See also
				Vol. 1 Sect. III-
				E-1
ign for optimus	None.	None.	•	A-3
Normance.				
sign for optimus				
	None.	Hone.	-	A-3
sign for optimus	None.	None.	-	A-3
riorpance.				
<u>.</u>				
tigo, materials	High-pressure hydrogen	Not clearly definable at this time.	Relevant data will be pro-	B.1
lection, development	environments from cryogenic		duced by Items 1.2.1, 1.2.3,	
testing.	to high temperatures; high-		1,2.4, and 1,3.1,	
	temperature oxidizing			
	environments, high merody-			
i. L	namic attesses, high-frequency fatigue, thermal			
	fatigue, severo thermal			
	sbock and thermal strenses.			
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Table VIII-1 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF NE

					Table VIII-1 (Continued)
			MATE	RIALS RESEARCH DEVELOPME	ENT AND TESTING NEEDED TO SUPP	ORT THE USE OF N
Item	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problems	Materials R,
3.	ROCKET PROPULSION ENGINES					
3.1.1 3.1.1 3.2.1 3.2.1	liyoʻrogen/oxygen.	H ₂ /O ₂	Turbine drive units for fuel pumps. Heat exchangers/gasifier (thrust chamber cooling).	Naterials selection.	High-pressure hydrogen or hydrogen/water environ- ments. Temperature range determined by design, but spans -423°F to about 1500°F. Principal problems at higher temperatures.	I termination of toughness, cree erties of candi steels, Ni allo high-pressure H environments at the range -428°
3.1.2			Long life combustion chamber and nozzle.	Bosign, materials selection, materials development and fabri- cation.	High-pressure, V. high-temerature, high-velocity $\rm H_2O$ environment. Thermal shock, thermal fatigue.	Materials and c
3.2.1	Monopropellant.	Hydrazine,	Very long life decemposition catalysts.	Catalyst research and devalogment.	Present iridium catalysts tend to lose activity and alumina catalyst support material deteriorates after long-term intermittent use.	Fundamental stumechanism of cadeterioration.
3.2.2						Development of catalyst.
4.	n.H.D. Systems					
4.1.1	Hydrogen fueled.	H ₂ /air or H ₂ /O ₂	H ₂ cooled magnets. Hot gas channel. Electrode materials.	Materials selection. Engineering design and development.	H ₂ environment effects. Righ-H ₂ C, high-temperature gas stream.	None at this ti
5.	INTERNAL COMBUSTION ENGINES					
5.1.1	Spark or diesel,	H ₂ /air	Fuel system components exposed to hydrogen.	Materials selection.	Possible H ₂ environment effects.	Long-term engin failure analysi
5.1.2			Ninor engine modifica- tions.	Engineering develop- ment.	None.	None.
3.2.1	Spark or diesel.	38 ₃	Fuel systems exposed to NH ₃ and H ₂ if fuel is partly dissociated to improve combustion.	Materials selection.	NH ₃ and H ₂ environment effects.	None.
5.2.2			Engine modifications and addition of NH ₃ dissociation equipment.	Engineering develop- ment,	None.	None,
5.3.1	Spark or diesel.	Hydrazine.	Not defined.	•	-	•
5.4.1	Spark or diesel.	%ethanol.	None.	-	-	-
					VIII-7	

Table VIII-1 (Coatinued) MARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF NEW FUELS

Solution	Materials Problems	Materials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. IV
solection.	High-pressure hydrogen or hydrogen/water environ- ments. Temperature range determined by design, but spans -423°F to about 1500°F. Principal problems at higher temperatures.	Determination of tensile, fracture toughness, creep and fatigue properties of candidate stainless steels, Ni alloys, Co alloys in high-pressure H ₂ and H ₂ + H ₂ O environments at temperatures in the range -428°F to 1500°F.	Overlaps with item 1.3.1 Continuation and extension of current programs.	B-2.`, B-3
materials materials mat and fabri-	High-pressure, V. high-temerature, high-velocity $\rm H_2O$ environment. Thermal shock, thermal fatigue.	Materials and component rig testing.	Possible long-term applica- tion to Item 1.4.3.	B-2.1
research and	Present iridium catalysts tend to lose activity and alumina catalyst support material deteriorates after long-term intermittent use.	Fundamental studies to elucidate mechanism of catalyst and substrate deterioration.	For small hydrazine decomposition control and accessory engines. No major problems with large hydrazine/oxidizer engines.	B-2.2
		Development of improved mixed-metal catalyst.	-	B-2.2
·				
selection. Ing design and ent.	H ₂ environment effects. High-H ₂ O, high-temperature gas stroam.	None at this time.	Clean H ₂ fuel superior to fossil fuels for large systems. Initiation of materials R, D, and T should await further engineering design.	С
selection.	Possible H ₂ environment effects.	Long-term engine tests, and materials farlure analysis.	No major difficulties in use of H_2 fuel expected.	D-2, D-5
ring develop-	None.	None.	<u>.</u>	D-2, D-5
soloction.	NH ₂ and H ₂ environment effects.	None.	NH, is a poor fuel for I.C engines, especially for diesels.	D-3, D-5
ing develop-	None.	None.	•	D-3, D-5
A Company A Property Company A P	•	•		D-4, D-5
Are El. change	-	-	Existing practice.	D-1, D-5
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Table VIII-1 (Continued)

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					Table VIII-1 (Continue	d)
			MATE	RIALS RESFARCH DEVELOR	PMENT AND TESTING NEEDED TO SUP	PORT THE USE OF NEW
Item No.	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problems	Materials R, I
6.	EXTERNAL COMBUSTION ENGINES					
6.1.1	Steam, recipro- cating.	All new fuels.	Fuel supply system and burners.	Design, materials selection.	No major problems.	None.
6.2.1	Stirling cycle (heating system).	All new fuels.	Fuel supply system and burners.	Design; materials	No major problems.	None.
6.3.1	Stirling cycle (H ₂ working fluid).	All new fuels,	Heater tubes, heat exchanger, operate at high temperatures with high internal H ₂ pressures. Working chambers must contain H ₂ with minimum loss. High H ₂ reservo'r pressures.	Materials develop- ment, selection.	High pressure (5000 psi) H ₂ environment effects from subzero temperatures to 1500°F on heat-resisting alloys.	Determination of toughness, creep, proporties of can steels, Ni alloys 5000 psi hydrogen subzero to 1500°F
6.3.2					Coatings to reduce hydrogen permeation through working chamber and heat exchanger tubing at high temperatures. Thermal shock and thermal fatigue.	Determination of for candidate all and 1500°F.
%					•	Develop coatings H_2 permeation at with substrate al stand thermal fat
	SPACE AND WATER HEATING					
7.1.1	Open flame combus- tion.	H ₂ /air.	Fuel supply system and burners.	Design; materials selection.	No major problem .	None.
7.2.1	Catalytic combus- tion.	H ₂ /air.	Catalytic burner to operate efficiently at low temperatures for long times.	Improved catalyst.	Low-cost, long-life, high- activity catalyst system that is resistant to poisoning by contaminants.	Fundamental studi- catalytic oxidati- transition metal- with empirical de- improved H ₂ oxida
17.3.1	Open flame combus- tion.	NH ₃ (disson- iated).	Efficient, low-cost NH ₃ dissociation equipment.	Improved catalyst.	Low-cost, long-life, high-activity catalyst system.	Development of aciation catalysts,
7.4.1	Catalytic combus- tion.	NH ₃	Catalytic burner to operate efficiently at low temperatures for long times.	Improved catalyst.	Low-cost, long-life, high activity oxidation catalyst that favors N ₂ and H ₂ O oxidation products.	Development of im
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	Table VIII-1 (Continued)		
<u>&</u> .	ENT AND TESTING NEEDED TO SUPP	ORT THE USE OF NEW FUELS		
				Report Reference
of Solution	Materials Problems	Materials R, D, and T Needs	Pemarks	in Vol. 2, Sect. IV
A-Southern And The An				
materials	No major problems.	None.	Careful attention to materials selection for H ₂ finel systems will be needed.	E-1, E-4
materials On.	No major problems.	None.	Careful attention to materials selection for ${\rm H_2}$ fuel systems will be needed.	R-2, E-4
is develop- blection.	High pressure (5000 psi) H ₂ environment effects from subzero temperatures to 1500°F on heat-resisting alloys.	Determination of tensile, fracture toughness, creep, and fatlgue properties of candidate stainless steels, Ni alloys, Co alloys in 5000 psi hydrogen at temperature from subzero to 1500°F.	Range of materials operating conditions covered by Items 1.3.1 and 3.1.1.	E-2, 6-4
કનન તત્વનો સાથા ત્યાં કર્યા હતું. ત્યાં માર્ચ મા	Coatings to reduce hydrogen permeation through working chamber and heat exchanger tubing at high temperatures. The mal shock and thermal fatigue.	Determination of $\rm H_2$ permeability data for candidate alloys up to 5000 psi and $1560^{\circ}7$.	-	E-2, E-4
		Develop coatings that are resistant to $\rm H_2$ permeation at 1500°F, compatible with substrate alloy and will withstand thermal fatigue.	-	E-2, E-4
materials	No major problems.	None.	-	F-1.1, F-6
catalyst.	Low-cost, long-life, high- activity catalyst system that is resistant to poisoning by conteminants.	Fundamental studies of mechanism of catalytic oxidation of H_2 , e.g., by transition metal carbides, combined with empirical development of improved H_2 oxidation catalysts.	-	F-3.1, S-2.3, F-6
catalyst.	Low-cost, long-life, high- activity catalyst system.	Development of low-cost NH ₃ dissociation catalysts.	Rolates to use of Nd ₂ in fuel ceils. This work would probably also relate to the development of improved NH ₃ synthesis catalysts.	F-2.1. F-2.3, F-6
Catalyst.	Low-cost, long-life, high activity oxidation catalyst that favors N ₂ and H ₂ O oxidation products.	Development of improved catalyst.	-	F-2.2, F-2.3, F-6
Catalyst. Catalyst. Catalyst.	VIII-8			

Table VIII-1 (Continueu)

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Item No.						
a A					Table VIII-1 (Continu	su)
			MAT	CERTALS RESEARCH DEVELOP	HENT AND TESTING NEEDED TO SU	
Item						
No.	Equipment Clas	s Fuel	Problem Area	Type of Solution	Materials Problems	Enterials R.
7.	SPACE AND WATER HEATING (Con't)					
7.5.1	Open flame combus- tion.	CO, Methanol.	None .	•	None.	None .
7.6.1	Catalytic combus- tion.	CO, Methanol.	Catalytic burners to operate efficiently at low temperatures for long times.	Improved catalysts.	low-cost, long-life, high- antivity catalyst systems.	Development of ba
	FUIL CELLS					· · · · · · · · · · · · · · · · · · ·
8.1.1 8.1.2 8.1.3 8.1.3	Alkaline electro- iyte.	H ₂ /sir (or O ₂).	Anode catalyst.	Catalyst dove opment.	Physical scanility of non- noble amone catalysts.	Nonnoble anode caphysical stabilit
8.1.2			Cathode catalyst.	Catalyst development,	Corresion of nonnoble cathode catalysts.	Nonnoble cathode polarization, low and improved phys
8.1.3						Nobie metal catal activity and stab catalyst leadings
8.1.4			Electrode structure.	Materials development.	Wetting and instability of polymer-bonded electrodes at 150°C and above.	Hydrophobic polyn materials for For 150°C.
8.1.5					Fabrication of gas-diffusion electrode structures with controlled porosity.	Improved controll materials for cle
8.1.6			Electrolyte matrix.	Materials development.	Present asbestos matrix can- not operate above 100°C.	- Low-cost matrix m operate at 150°C.
8.1.7						Improved ion-exch
8.1.6 8.1.7 8.2.2 8.2.3	Acid electrolyte.	H ₂ /air (or O ₂),	Cathode catalyst.	Catalyst development.	Relatively low activity, high polarization of platinum catalyst for O ₂ reduction in acid electrolyte.	Improved noble me improved activity stability at low
8.2.2						Low-cost, nonnobl with moderate act stability and cor
			Electrolyte matrix.	Materials development.	Less expensive, more robust matrix materials.	Improved ion-exch
8.3.1	holten carbonate electrolyte.	H ₂ /air (or O ₂).	Anode and cathode catalysts.	Materials development.	Lack of catalyst stability.	Anode and cathode with improved sta
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	Table VIII-1 (Continue			
rksearch develop:	ENT AND TESTING NEEDED TO SUP	PORT THE USE OF NEW FUELS		
of Solution	Materials Problems	Materials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. IV
		_	_	
	None.	None.	-	F-4, F-5, F-6
ed catalysts.	Low-cost, long-life, high- sciivity catalyst systems.	Development of base-metal catalysts.	_	F-4, F-5, F-6
		•		
st development.	Physical stability of non- noble anode catalysis,	Nonnoble anode catalyst with improved physical stability at 150°C and above.	Relatively low corrosivity of alkaline elect ciyts permits wide choice of materials of construction.	G-3.1.1, G-3.1.6, G-3,9.1
az development.	Corrosion of nonnoble cathode catalysts.	Nonnoble cathode catalyst with lower polarization, lower corrosion rate and improved physical stability.	•	i-3.1.3, G-3.1.6, G-3.9.1
		Noble metal catalysts with high activity and stability at lower catalyst loadings.	-	G-3.1.2, G-3.1.6, G-3.9.1
Lals development.	Wetting and instability of polymer-bunded electrodes at 150°C and above.	Eydrophobic polymeric bonding materials for service at above 150°C.	•	G-3.1.3, G-3.1.6 G-3.9.2
	Fabrication of gas-diffusion elections structures with controlled porosity.	improved controlled-porosity materials for electrode attructures.	-	G-3.9.2
ezs development.	Present asbastos matrix can- not operate above 100°C.	Low-cost matrix materials able to operate at 150°C.	•	G-3.1.4, G-3.1.6 G-3.9.3
		Improved ion-exchange membranes.	•	
nt developmen.	Polatively low activity, high polarization of platinum catalyst for \mathcal{O}_2 reduction in acid electrolyto.	Low-cost matrix materials able to operate at 150°C. Improved ion-exchange membranes. Improved noble metal catalyst with improved activity and physical stability at low catalyst loadings. Low-cost, nonnoble metal catalysts with moderate activity, and high stability and corrosion resistance. Improved ion-exchange membranes.	Acid electrolyte limits choice of materials of construction.	G-3.2.2, G-3.2.6, G-3.9.1
		Low-cost, nonnoble metal catalysts with moderate activity, and high stability and corrosion resistance.	-	G-3.2.2, G-3.2.6 G-3.9.1
als development.	Less expensive, more robust matrix materials.	Improved ion-exchange membranes.	-	G-3.2.6, G-3.9.3
ials devsiopment.	Lack of ca'llyst stability.	Anode and cathode catalysts with improved stability.	Noiten carbonate cells require use of \mathbb{CO}_2 addition to cathode reactant (air).	G-3.3.1, G-3.3.2, G-3.3.6, G-3.9.1
	6-111A			

TABLE VIII-1 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF 1

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V. Silk Line						
					TABLE VIII-1 (Continu	ed)
- -			МА	TERIALS RESEARCH DEVELOP	MENT AND TESTING NEEDED TO SU	PPORT THE USE OF 1
Item No.	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problems	Materials R,
8.	FUEL CELLS (Con't)					
8.3.	2 Molten carbonate (Con't)	H ₂ /2ir (or O ₂)	Electrode structure.	Materials development.	Need for ceramic structure analogous to hydrophobic polymer-bonded systems for low-temperature cells.	Ceramics which control of wetting angle
8.3.	3		Electrolyte matrix.	Materials development.	Thermal cracking of the ceramic electrolyte matrix tile.	Ceramic matrix material controlled porosi resistance to the
Item 8. 8.3. 8.4. 8.5. 8.5.	l inorganic solid electrolyte.	H ₂ /air (or O ₂).	High operating temperature.	Katerials development and design.	Differential expansion between cell components, low conductivity of elec- trolyte, contact between matrices and electrodes head and gas leakage.	Development of comatching expansion thin, stable, high matrix materials.
8.4.	2			Basic materials research.	Inorganic solid electrolyte operating at lower temperatures.	Study of low-temp conducting inorga
8.5.	l Direct methanol fuel cells.	Methanol.	Low activity of methanol electrode.	Catalyst development.	Anode catalyst.	Higher activity a
8.5.	2		Cross-over of methanol to cathode.	Materials development.	Methanol dissolved in elec- trolyte diffuses through matrix.	Matrix material (to methanol.
8.5.	3				Air electrode structure weited by cross-over methanol.	Electrode structu can control wetti
8.5.	4			Catalyst development.	Platinum cathode catalyst also catalyses direct oxidation of cross-over methanol.	Selective cathode more oxygen reduc methanol oxidatio
8.5. 8.6. 8.7. 8.7. 8.7.	l Indirect methanol fuel cells.	Reformed methanol (H ₂ , CO, CO ₂).	Same as hydrogen fuel cells.	-	-	_
8.7	l Hydrazine iuel cells.	Hydrazine.	Chemical decomposition of hydrazine at anode.	Catalyst development.	Anode catalyst may decompose hydrazine, oxidize hydrogen formed, cause NH ₃ evolution.	hydrazine oxidati
8.7	,2		Cross-over of hydra- zine to cathode.	Materials development.	Hydrazine dissolved in electrolyte diffuses through matrix.	Matrix material (: to hydrazine.
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	TABLE V./II-1 (Continu			
	TABLE V. / I-1 (Continue	ed)		
BEARCH DEVELOP	MENT AND TESTING NEEDED TO SU	PPORT THE USE OF NEW FUELS		
of Solution	Materials Problems	Materials R, D and T Needs	Remarks	Report Reference in Vol. 2, Sect. IV
development.	Need for ceramic structure analogous to hydrophobic polymer-bonded bystems for low-temperature cells.	Ceramics which can provide control of wetting angle in molten carbonate.	-	G-3.3.3, G-3.9.3
a development.	Thermal cracking of the ceramic electrolyte matrix tile.	Ceramic matrix materials with controlled porosity and improved resistance to thermal cycling.	-	G-3.3.4, G-3.9.3
development n. terials	Differential expansion between cell components, low conductivity of elec- trolyte, contact between matrices and electrodes head and gas leakage.	Development of cell materials with matching expansions, development of thin, stable, high-conductivity matrix materials.	Operating range 700° to 1000°C.	G-3.4, G-3.4.4
Kerials	Inorganic solid electrolyte operating at lower temperatures.	Study of low-temperature, ion-conducting inorganic solids.	Breakthrough analogous to discovery of 3-alumina needed.	G-3.4.3, G-3.4.4
development.	Anode catalyst.	Higher activity anode catalyst.	-	G-3.5, G-3.5.1
development.	Methanol dissolved in electrolyte diffuses through matrix.	Matrix material (separator) impervious to methanol.	-	G-3.5, G-3.5.1
s development.	Air electrode structure wetted by cross-over methanol.	Electrode structure materials that can control wetting angle.	-	G-3.5, G-3.5.1
dev elopment.	Platinum cathode catalyst also catalyses direct oxidation of cross-over methanol.	Selective cathode catalysts that promote oxygen reduction but not methanol oxidation.	-	G-3.5, G-3.5.1
	-	-	Reformed methanol can be used in acidic, molten carbonate or solid electrolyte fuel cell	
development.	hydrazine, oxidize hydrogen	Selective anode catalyst that promotes hydrazine oxidation but not its decomposition, or the oxidation of hydrogen.		G-3.6, G-3.9
development.	Hydrazine dissolved in electrolyte diffuses through matrix.	Matrix material (separator) impervious to hydrazine.	Equivalent to Item 8.5.2.	G-3.6, G-3.9
development.	VIII-10			

Table VIII-1 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF NE

8. FUE . C	quipment Class CELLS (Concluded	Fuel	MAZ	TERIALS RESEARCH DEVELOP	Table VIII-l (Continued	1)
Item No. Ec 8. FUE C 8.7.3 Hydraz cells.	CELLS (Concluded	Fuel	MAT	TERIALS RESEARCH DEVELOP	Table VIII-l (Continued	1)
Item No. Ec 8. FUE C 8.7.3 Hydraz cells.	CELLS (Concluded	Fuel	MAT	TERIALS RESEARCH DEVELOP	Table VIII-1 (Continuo	i)
Item No. Ec 8. FUE C 8.7.3 Hydraz cells.	CELLS (Concluded	Fuel	MAT	TERIALS RESEARCH DEVELOP		
1 tem No. Eq. (8. FUE (1. Cells.)	CELLS (Concluded	Fuel			MENT AND TESTING NEEDED TO SU	PPOKT THE USE OF NE
8. FUE (ine fuel		Problem Area	Type of Solution	Materials Problems	Materia's R, D,
8.7.3 Hydraz cells.		1)				
Š.	(con t)	Hydrazine	Hydrazine cross-over.	Catalyst development.	Cathode catalyst catalyses decomposition or oxidation of cross-over hydrazine.	Selective cathode motes oxygon reduc zine decomposition
8.8.1 Direct		NH ₃	Anode.	Catalyst research.	Anode activity decreases with time.	Basic studies to d
8.9.1 indire	ect NH ₃ fuel	NH ₃ (dissoci- ated to N ₂ and H ₂).	Same as hydrogen fuel cells.	-	~	-
8.10.1 Regene	erative fuel	$\mathrm{H_2/O_2}$	Variations of electrode potential as cell operation is reversed.	Catalyst development.	Decreased catalyst life.	Electrocatalysts t to potential cycli
8.10.2				Electrode structure development.	Dual function electrode structures.	Electrode structur
8.10.3					Dual electrode structures.	Two electrode stru appropriate cataly
8.11.1 Fuel of general	cells al.	Various	Polymeric materials for electrical insulators and seals for service above 150°C.	Design; materials and fabrication development.	Polymeric materials with long-term, high-temperature stability at e 150°C.	Polymeric material testing.
8.11.2			Reactant supply systems.	Materials selection.	${\rm H_2}$, ${\rm NH_3}$, hydrazine environments.	See item headings
8.11.3 8.11.4			'n. eory of electro- catalyst behavior.	Basic electrocatalysis.	Wechanism of electro- catalysis.	Studies of effects electrolytes, and a on electrocatalyst, face.
8.11.4			Electrocatalyst screening.	Measurements of single electrode characteristics.	Empirical selection of electrocatalysts.	Determination of dimaterials.
9. HIGH	ENERGIY DENSITY RIES					
9.1.1 Aqueon ally a	us, electric- rechargeable, r, Zn/O ₂ .	Zn	Zn electrode.	Design and engineering; basic research.	Shope change (dendrite growth) of Zn electrode,	Studies of mechanis growth.
9.1.1 Aqueot ally zn/ai:			Air electrode; poor charge/discharge cycle efficiency.	Catalyst development, electrode design.	High catalyst loadings required.	Dual function or to catalysts for oxyge oxygen reduction.
					VIII-11	
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All Market and the second seco				
Folkers Total				
호 	Table VIII-1 (Continued)		
ALS RESEARCH DEVELOPS	MENT AND TESTING NEEDED TO SU	PPORT THE USE (NEW FUELS		
po of Solution	Materials Problems	Materials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. I
alyst development.	Cathode catalyst catalyses decomposition or oxidation of cross-over hydrazine.	Selective cat! Die catalysts that promotes oxygen reduction but not hydrazine decomposition or oxidation.	-	G-3.6, G-3.9
, •	Anode activity decreases with time.	Basic studies to determine cause of decline in anode activity.	-	B-3,7
real state of the	_	<u>.</u>	Can be used in all ${\rm H_2}$ fuel cell types including alkaline.	G-3.7
salyst development.	Decreased catalyst life.	Electrocatalysts that are insensitive to potential cycling.	-	G-3.8, G-3.9
ctrode structure	Dual function electrode structures.	Electrode structures with two distinct catalysts.	-	G-3.8, G-3.9
to the extension in the later of the extension in the extension in the later of the extension in	Dual electrode structures.	Two electrode structures each with appropriate catalysis.	-	G-3.8, G-3.9
ign; materials and frication develop-	Polymeric materials with long-term, high-temperature stability above 150°C.	Polymeric materials development and testing.	Will probably be satisfied by improved commercial products.	G-3.9.4
erials selection.	$\rm H_2$, $\rm NH_{33}$ hydrazine environments.	See item headings 1 through 7.	Requirements common to other uses of new fuels.	G-3.9.4
ic electrocatalysis.	Mechanism of electro- catalysis.	Studies of effects of reactants, electrolytes, and applied potential on electrocatalyst/electrolyte interface.	-	G-3.9.4
surements of single etrode character- ics.	Empirical selection of electrocatalysts.	Determination of data for candidate materials.	-	G-3.9.4
ign and engineering;	Shape change (dendrite growth) of Zn electrode.	Studies of mechanism of dendrite growth.	Future of this battery depends on solving Zn electrode problem.	Н-3.1.1, Н-3.1.3
talyst development, ctrode design.	ligh catalyst loadings required.	Dual function or two separate catalysts for oxygen evolution and oxygen reduction.	-	н-3.1.3

Table VIII-1 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF

No.	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problem	Materials R,
9.	HIGH ENERGY DENSITY BATTERIES (Continued	1)				
9.2.1	Aqueous, mechanic- ally rechargeable, Zn/air.	Zn	Electrolyte leakage.	Improved design, seal materials.	Seal materials.	Support to engi
9.3.1	Aqueous, Zn/Ni.	Zn	Zn electrode.	See Item 9.1.1.	See Item 9.1.1.	See 1tem 9.1.1.
9.4.1	Aqueous Zn/Cl ₂ .	Zn	Zn electrode.	See Item 9.1.1.	See Item 9.1.1.	See Item 9.1.1.
9.4.2			Cl, electrode.	Materials development.	Polarization of Cl ₂ electrode.	More active Cl ₂
9.5.1	Aqueous Li/Ni.	Lı	Li electrode; shelf life.	Materials development, corrosion research.	Li corrosion reaction in the absence of external current flow.	Alloying of Li to limit Li cor
9.5.2			${\rm H_2}$ evolved, must be vented.	Materials selection.	H ₂ environment effects.	None.
9.6.1	Aprotic solvent Li batteries.	Li	Li electrode; low power density, shelf life, slow start-up due to electronic passivation.	Electrochemical studies, battery design.	Electrode/electrolyte interactions.	Support to elec
9.7.1	Molten salt, high temperature, Li/Cl ₂ battery.	Li	Li electrode.	Electrode material development.	Dissolution and migration of lithium.	Development of maximum cell vo
9.7.2			Cl ₂ electrode; low current capacity.	Electrode and electrolyte development.	Electrode and electrolyte materials.	Support for ele development stu
9.7.3			Materials of construction.	Materials development and selection.	Attack of materials by molten Li.	Ceramics and co
9.7.4					Corrosion by molten alkali- halide electrolyte.	Alloys and coat molten halides.
9.7.5					Electrical feed-throughs,	Containment mat brazing alloys coefficients.
9.7.6					Containment of high temperature Cl ₂ gas.	Alloys and coat Cl_2 gas.
9.8.1	Li/S molton salt, battery.	Li	Sulfur electrode.	Materials development and selection.	Attack of sulfur on materials of construction.	Materials compa

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	Table Vill-l (Continued MERT AND TESTING NEEDED TO SU Materials Problem			
2 V				
e e e e e e e e e e e e e e e e e e e	Table VIII-1 (Continued)		
LALS RESEARCH DEVELOP	MENT AND TESTING NEEDED TO SU	PPORT THE USE OF NEW FUELS		
distribution of the state of th				
to of Solution	Materials Problem	Materials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. 1
				24 103. 21 500.
roved design, seal prials.	Scal materials.	Support to engineering development.	-	H-3.1.1
erials.				
	0.2.2.			
Item 9.1.1.	See Item 9.1.1.	See Item 9.1.1.	Electrically rechargeable.	H-3.1.4
Ztem 9.1.1.	See Item 9.1.1.	See Item 9.1.1.	Electrically rechargeable	H-3.1.5
rials development.	Polarization of Ci ₂ electrode.	More active Cl_2 electrode substrate.	Would improve power density.	H-3.1.5
	• • • • • • • • • • • • • • • • • • • •		Dulana and a second and a second	
rials development,	Li corrosion reaction in the absence of external current flow.	Alloying of Li or use of inhibitors to limit Li corrosion.	Primary or mechanically rechargeable battery.	н-3.1.6
rials selection.	H ₂ environment effects.	None.	-	н-3.1.6
trochemical lies, battery	Llectrode/electrolyte interactions.	Support to electrochemical and design studies.	No $\rm H_2$ evolved. Can be sealed for use as primary battery.	н-3.1
trode material	Dissolution and migration of lithium.	Development of solid Li alloys giving maximum cell voltage.	Operates from 350° to 650°C depending on electrolyte composition.	н-3.3.1
trode and elec-	Electrode and electrolyte materials.	Support for electrode and electrolyte development studies.	-	н-3.3,1
cials development	Attack of materials by molten Li.	Ceramics and coatings resistant to molten Li.	-	н-3.3.1
e constant of the constant of	Corrosion by molton alkali- halide electrolyte.	Alloys and coatings resistant to moiten halides.	-	н-3.3.1
	Electrical feed-throughs.	Containment materials, insulators and brazing alloys with matching expansion coefficients.	-	н-3.3.1
	Containment of high temperature Cl_2 gas.	Alloys and coatings resistant to hot CI ₂ gas.	•	н-3.3.1
ials development election,	Attack of sulfur on materials of construction.	Materials compatibility testing or electrode materials modification.	Operates at 375° to 400°C.	н-3.3.1

Table VIII-1 (Concluded)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE USE OF

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			MA	TERIALS RESEARCH DEVELOP	Table VIII-1 (Concluded)	
Item No.	Equipment Class	Fuel	Problem Area	Type of Solution	Materials Problem	Mcterials f
9.	HIGH ENERGY DENSITY BATTERIES (Concluded	1)				
9.9.1	Molten sait, low-temperature A1/Cl ₂ battery.	A1	Waterials of construction.	Naterials selection.	None.	None.
3.10.1	Solid electrolyte Na/S battery.	Na	3-alumina electrolyte.	Materials research; ceramic processing development.	Improved Na-ion conductivity, resistance to intergranular attack, to thermal cracking and mechanical strength.	Modification an alumina composi
9.10.2						Investigation o
9.11.1	General.	Various.	Materials of construction.	Battery design and development, materials development and selec- tion.	Materials compatibility with electrodes and elec- trolytes; seals, feed- throughs, insulators, etc.	Supporting mate development stu battery F, J, a
9.11.2			Solid electrolytes.	Materials research and development.	Improved solid electrolytes and cation/anion exchange membranes.	Search for new ducting solids.
						
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	Tabie VIII-1 (Concluded)			
MEARCH DEVELOPS	MENT AND TESTING NEEDED TO SU	IPPORT THE USE OF NEW FUELS		
Jolution				Report Reference
Colution	Materials Problem	Materials R. D. and T Needs	Remarks	in Vol. 2, Sect. IV
selection.	None.	None.	Operates at 60° to 150°C.	н-3.3.2
gresearch;	Improved Na-ton conduc-	Modification and optimization of g-	Operates at about 300°C.	H-3.4
rocessing	tivity, resistance to	siumina composition.	•	
2	intergranular attack, to thermal cracking and			
	mechanical strength.			
and the state of t		Investigation of alternative Na-ion conductors.	-	н-3.4
esign and at, materials	Marerials compatibility with electrodes and electrodes	Surporting materials engineering and development studies as part of	-	H-3.1
at and selec-	tro ytes; seals, feed- tr oughs, insulators, etc.	battery R, D, and T programs.		
research and		Search for new low-temperature ion-con-	-	H-3.5
escarch and	and cation/anion exchange membranes.	ducting solids.		
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TABLE VIII-2 MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE PRODUCTION OF NEW FUELS

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE PRODUCT

Table VIII-2

Item No.	Process	Fuel Produced	Problem Area	Type of Solution	Materials Problem	Matericl
1.	Conventional Electrolyzers					
1 1.1	Unipolar tank type.	H2 (02)	Efficiency and cost.	Advanced electrolyzers.	No major problems.	None.
1.2.1	Bipolar, filter press type.	H ₂ (O ₂)	Efficiency and cost.	Advanced electrolyzers.	No major problems.	None.
2.	ADVANCED ELECTROLYZERS					
2.1.1	Alkaline electrolyte,	H ₂ (O ₂)	Higher temperature operation to increase	Materials development.	Corrosion of positive (oxygen) electrode.	Materials wit superior to n
2.1.2			efficiency.		Present diaphragm material (asbestos) limited to 100°C	Higher temper (200°C desira
2.1.3					Present frame materials (polysulphones) limited to 150°C.	Higher temper materials, an
2.1.4			Reduction of over- potentials.	Materials development and design.	Relatively low conductivity and bulky diaphragm materials.	High-conducti higher temper diaphragm (ma
2.1.5					Effective area of electrodes.	High effectiv materials.
2.1.6				Catalyst development.	Electrode activity.	Improved nonn catalysts.
2.2.1						
2.2.1	Solid polymer electrolyte.	H ₂ (O ₂)	SPE membrane.	Materials research and development.	Present SPE membrane is expensive, temperature limited (125°C) and somewhat lacking in mechanical and chemical stability.	Development o perature SPE (anion or cat improved mech bility.
2.2.2			Electrodes.	Catalyst development.	High loadings of noble metal catalysts.	Nonnoble meta
2.2.3				Engineering and materials development.	Electrode/membrane contact resistance.	Support for e
2.3.1	Inorganic solid electrolyte.	H ₂ (O ₂)	Very high operating temperature (1000°C).	Materials R and D.	Solid inorganic electrolyte membranes with good ionic conductivity below 400°C.	Basic studies erties of cer
2.3.2			Materials and engi- neering development.	Materials and engi- neering development.	Materials of construction,	Materials wor neering devel

Table VIII-2 **BEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE PRODUCTION OF NEW FUELS

pe of Solution	Materials Problem	Materials R, D, and T Needs	Remarks	Report References in Vol. 2, Sect. V
iced electrolyzers.	No major problems.	None.	Established technology.	A-2.1.1
sed electrolyzers.	No major problems.	None.	Established technology.	A-2.1.2
mals development.		Materials with corrosion resistance superior to nickel.	Nickel is now used.	A-2.2.1, A-4
	(oxygen) electrode. Present diaphragm material (asbestos) limited to 100°C	Higher temperature diaphragm materials	-	A-2.2.1, A-4
Server Annual Control of the Control	Present frame materials (polysulphones) limited to 150°C.	Higher temperature, low-cost frame materials, and methods of fabrication.	-	A-2.2.1, A-4
dals development beign.	Relatively low conductivity and bulky diaphragm materials.	Hagh-conductivity, compact, (and higher temperatures - Item 2.1.2) diaphragm (matrix) materials.	Less bulky diaphragm would permit closer electrode spacing and less voltage drop in the electrolyte.	A-2.2.1, A-4
	Effective area of electrodes.	High effective surface area electrode materials.	-	A~2.2.1, A-4
lyst development.	Electrode activity.	Improved nonnoble, low-cost electro-catalysts.	Ni and Fe electrode materials now used are moderately effective.	A-2.2.1, A-4
rials research and comment.	Present SPE membrane is expensive, temperature limited (125°C) and somewhat lacking in mechanical and chemical stability.	Development of low-cost, higher tem- perature SPE membranes with high ionic (anion or cation) conductivity and improved mechanical and chemical sta- bility.	-	A-2.2.2, A-4
lyst development.	High loadings of noble metal catalysts.	Nonnoble metal catalysts.	Noble metal catalysts probably acceptable for DoD uses, but probably not for large-scale industrial use.	A-2.2.2, A-4
meering and Fials development.	Electrode/membrane contact resistance.	Support for engineering development.	-	A-2.2.2, A-4
rials P. and D.	Solid inorganic electrolyte membranes with good ionic conductivity below 400°C.	Basic studies of ion-conducting properties of ceramics.	High-temperature electro- lyzers have limited actual thermal efficiencies.	A-2.2.3, A-4
rials and engi- ing development.	Materials of construction.	Materials work in support of engineering development.	Problems similar to those of high-temperature fuel cells (Item 8.4.1 in Table VIII-1).	A-2.2.3
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Table VIII-2 (Continued)

MATERIALS RESEARCH DEVE OPMENT AND TESTING NEEDED TO SUPPORT THE PROD

Item No.	Process	Fuel Produced	Problem Area	Type of Solution	Materials Problem	Mater:
2.	ADVANCED ELECTROLYZE	ERS				
2.4.1	Acid electrolyte.	H ₂ (O ₂)	Oxygen electrode.	Materials Jelection.	Corrosion of oxygen electrode materials and associated conductors.	Materials sel
2.5.1	General.	H ₂ (O ₂)	Naterials resources availability.	Technoeconomic analysis of materials supply needs for competing electrolyzer technologies.	-	None.
2.5.2			Materials of construction, auxilliary equipment.		$\mathrm{H_2}$ environment and $\mathrm{O_2}$ environment effects.	Materials terelectrolyzer of temperature
3.	THERMOCHEMICAL SPLITTING					
3.1.1	Several multistep cycles (all are now at a conceptual stage).	H ₂ (O ₂)	Materials of construction.	Materials development and selection; engi- neering design and development.	High-temperature corrosion of metallic and ceramic materials in halogens, halides, metal oxides, liquid metal, hydrogen, oxygen, steam.	Naterials and possible need materials and
4.	HYDROGEN LIQUE- FACTION					
4.1.1	Compression lique-faction of ${\rm H_2}$ gas.	Liquid H ₂	Compressor development.	Engineering design and development; materials develor ant and selection.	Materials for use as cylinder liners, pistons, piston rings, seals, bearings, turbine blading in advanced compressors at ambient and cryogenic temperatures in H ₂ environment.	H ₂ -resistant, weight mater:
4.1.2						Low friction binations for
4.1.3						Seal material

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() () ()	Table VIII-2 (Continued)									
DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE PRODUCTION OF NEW FUELS										
lution	Materials Problem	Naterials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. V						
		total and a stock of the stock								
6 6	· · · · · · · · · · · · · · · · · · ·									
etion.	Corrosion of oxygen elec-	Materials selection and small-scale	Acid electrolytes reject CO ₂	A-2.2.4						
でを言う	trode materials and associ- ated conductors.	operational testing.	and are of interest for use in life support systems.							
c analysis	-	None.	Essential background for	A-4						
analysis supply seting			policy decis lons.							
tech-										
ection.	H ₂ environment and O ₂	Materials testing under	Information required will be	A-4						
એક કુલ કુલ કુલ કુલ કુલ કુલ કુલ કુલ કુલ કુ	environment effects.	electrolyzer operating conditions of temperature and pressure.	provided by work relating to Section IV of this report.							
*			(See Table VIII-1.)							
elopment engi- and	High-teaperature corrosion of metallic and ceramic	Materials and component testing; possible need for corrosion-resistant	Specific programs can be planned when further	B-1, B-2, B-3, B-4						
and	materials in halogens, halides, metal oxides,	materials and coatings development.	feasibility studies have been made and prototype process							
	liquid metal, hydrogen, oxygen, steam.		cycles have been selected.							
sign and	Materials for use as	H2-resistant, high-strength light-	Materials support for engi-	c						
materials md selec-	cylinder liners, pistons, piston rings, seals,	weight materials.	neering dev opment.							
	bearings, turbine blading in advanced compressors									
	st ambient and cryogenic temperatures in H ₂									
	environment.	Low friction and wear materials com-	_	c						
		binations for service in H ₂ .	-	•						
		Seal materials, bearing materials, low-expansion alloys.	-	c						
	Naterials for use as cylinder liners, pistons, piston rings, seals, bearings, turbine blading in advanced compressors at ambient and cryogenic temperatures in H ₂ environment.									
	VIII-15									
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Table VIII-2 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE

Item		Fuel				
No.	Process	P Produced	Froblem Area	Type of Solution	Materials Problem	Ma ⁻
5.	AMMONIA PRODUCTION	·				
5.1.1	Modified Haber Process.	NH ₃	Pressure vessels.	Materials selection and design.	Effects of long time expo- sure of pressure vessel materials to high pressure H ₂ , N ₂ , NH ₃ at moderate temperatures.	Probably
5.1.2					Improved on-site and in- service nondestructive testing techniques.	Probably
5.1.3				Alternative designs and materials, e.g. metal lined, pre- stressed or fiber reinforced cements or concrete vessels.	Materials for very large pressure vessels.	Developme reinforce material:
5.1.4			Ancillary equipment, compressors.	Materials selection.	${\rm H_2}$ environment effects.	Naterial: high pres
5.1.5			NH ₃ condensation and storage.	Materials research; materials selection.	Stress corrosion cracking of quenched and tempered steels in liquid ammonia.	Material: mechanism of SCC in testing of material:
6.	Hydrazine Production					
6.1.1	Modified Raschig process.	Hydrazine	Possible problems in scaled-up production.	Materials selection;	•	None at 1
7.	PRODUCTION OF BORANES AND SILANES		The pro-			
7.1.1	Verious smell	Boranes and	Pessible problems in scaled-up production.	Materials selection; engineering.	•	None at 1

	Table 1877 0 (0.14)			
ARCH DEVELOPAS	Table VIII-2 (Contine and Testing Needed to sup	PORT THE PRODUCTION OF NEW FUZLS		
of Solution	Material. Problem	Materials R, D, and T Needs	Remarks	Report Reference in Vol. 2, Sect. V
\$-				
selection	Effects of long time exposure of pressure vessel materials to high pressure H ₂ , N ₂ , NH ₃ at moderate temperatures.	Probably no special programs required.	Necessary test data will be collected from related industrial experience and test programs.	D-1.1.1
	Improved on-site and in- service nondestructive testing techniques.	Probably no special programs required.	Appropriate NDT and E techniques will be developed for nuclear power plants.	D-1.1.1
ve designs ials, e.g. id, pre- or fiber d cements te vessels.	Materials for very large pressure vessels.	Development and testing of fiber- reinforced cement and concrete materials and structures.	Likely to be of broad general application.	D-1.1.1
selection.	H ₂ environment effects.	Materials and component testing in high pressure ${\rm H_2}.$	Would profit from advanced compressor development. Items 4.1.1, 4.1.2, 4.1.3.	D-1.1.2
research; selection.	Stress corrosion cracking of quenched and tempered steels in liquid ammonia.	Materials research to establish mechanism and limiting conditions of SCC in liquid ammonia and testing of candiate SCC-resistant materials.	See Volume 1, Section III-D-1, same as Item 1.5.1 in Table VIII-1.	D-1,1.2
relection;	-	None at this time.	Reexamine if large-scale production becomes likely,	D-2
salection;	-	None at this time.	Reexamine if large-scale production becomes likely.	D-2
	VIII-16			

Table VIII-2 (Concluded)

MATERIALS RESEARCH DEVELOPMENT AND TESTING WESDED TO SUPPORT THE PRO

1tem		Fuel				
No.	Process	Produced	Problem Area	Type of Solution	Materials Problem	Mate:
8.	PARTIALLY OXYGENATED FUELS					
8.1.1	H ₂ production.	H ₂ inter- mediate	See Items 1, 2, and 3.	See Items 1, 2, and 3.	See Items 1, 2, and 3.	See Items 1
8.2.1	CO ₂ production from the air.	CO ₂ inter- mediate	Advances in compressor and CO_2 adsorption methods.	Engineering design and development. Process development.	None.	None.
8.2.2	CO_2 production from limestone.	CO ₂ intermediate	Mining and materials handling (nuclear reactor problems excluded).	Engineering and materials development.	Low-cost materials with high abrasion resistance.	Probably nor
8.3.1	Co production.	co	No existing production process.	Process development; catalyst selection.	Effective, long-life catalyst.	Catalyst dev
8.4.1	Methanol production.	Nethanol	None.	•	-	None.
8.5.1	Methanol production (electrolytic).	Methanol	Conceptual process.	Process development; catalyst development.	Electrocatalyst research and development.	Development methanol ele

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		Materials R, D, and T Needs See Items 1, 2, and 3.		····
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CEVELOPAL	ENT AND TESTING REEDED TO SU	PPORT THE PRODUCTION OF YAW FUELS		
				Report Reference
Solution	Materials Problem	Materials R, D, and T Needs	Remarks	in Vol. 2 Sect. V
in the				
2, and 3.	See Items 1, 2, and 3.	See Items 1, 2, and 3.	${\rm H_2}$ required for production	E, A, B,
			of CO and methanol.	
lesign and	None.	None.	CO2 required for production	E-1
Process			of CO ₂ and methanol.	
and relopment.			* • • • • • • • • • • • • • • • • • • •	
elopment.	Low-cost materials with high abrasion resistance.	Probably none required.	Calcination using nuclear heat. Improved, low-cost	E-2
	<u> </u>		abrasicn-resistant	
			materials provided by industry development,	
				<u></u>
topment;	Effective, long-life	Catalyst development.	Reversed shift reaction.	E-3
ection.	catalyst.			
	-	Nc ne .	Existing technology	E-4.1
	Electrocatalyst research and development.	Development of electrocatalysts for	Mat rials of construction of plant similar to Item 2.1.	E-4.2
elopment.	and development.	methanol electrode.	of prant similar to from 2.1.	·
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TABLE VIII-3 MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE TRANSPORTATION OF NEW FUELS

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE TRANSPORT.

Table VIII-3

Item	Mode of Transportation	Fuel	Problem Area	Type of Solution	Materials Problem	Mater
1.	EXISTING PIPELINES					
1.1.1	H ₂ pipelines.	H ₂ gas.	Acceptability of exist- ing pipelines for H ₂ transmission and dis- tribution.	Materials testing.	H ₂ environment effects over long times on mechanical properties of material and welds of existing pipelines.	line mater hydrogen a minor quan and illumi from -60° unnotched toughness, static loa
1.1.2				Full-scale pipe testing.	H ₂ environment effects over long times on service behavior of existing pipe- lines.	Full-scale and used p taminated tures and tuating) c service an
1.1.3				Basic research.	Role of surfaces in adsorption and dissociation of $\rm H_2$.	Study of e sulfide, a on mechani
1.2.1	NH ₃ pipelines.	NH ₃ liquid.	Acceptability of existing pipelines for transmission and distribution of NH ₃ .	Materials testing.	Stress corrosion cracking of steels in liquid NH ₃ .	Long-term new and us in NH ₃ as structure stress, no and NH ₃ pr
1.2.2				Full-scale pipe testing,	Stress corrosion cracking of steels in liquid NH ₃ .	Full-scale pipelines NH ₃ enviro pressures service an
1,2,3				Basic research.	Stress corrosion cracking of steels in liquid NH ₃ .	Studies to problem of liquid NH ₃
1.3.1	CO pipelines.	со	Acceptability of existing pipelines for transmission and dis- tribution of CO.	Materials testing.	Possible environmental effects of CO at high pressures for long times on pipeline materials.	long-term line steel deteriorat properties contaminat
1.4.1	Menthanol pipelines.	Methanol.	None.	•	None.	None.
2.	NEW PIPELINES					
2.1.1	H ₂ pipelines.	H ₂ gas.	Acceptability of new metalite pipeline materials and installations for H ₂ service.	Waterials testing and selection.	H ₂ environment effects over long times on higher strength pipeline steels.	Long-term line steel H ₂ contami of O ₂ , H ₂ (Test data 160°F on r strength, failure ur fatigue.

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Table VIII-3 TELOPMENT AND TESTING NEEDED TO SUPPORT THE TRANSPORTATION OF NEW FUELS

S olution	Materials Problem	Materials R, D, and T Needs	Remarks	Report Reference Vol. 2, Sect. VI
testing.	H ₂ environment effects over long times on mechanical properties of material and welds of existing pipelines.	Long-term testing of new and used pipeline materials and welds in pure hydrogen and hydrogen contaminated with minor quantities of O_2 , H_2O , odorants and illuminants. Test data required from -60° to $+160^\circ$ F on notched and unnotched tensile strength, fracture toughness, delayed failure under static load and low cycle fatigue.	from Item 1.1.2 will permit	A-4, A-6
pipe	H ₂ environment effects over long times on service behavior of existing pipe- lines.	Full-scale testing of sections of new and used pipelines in pure and contaminated \mathbb{H}_2 environments at temperatures and pressures (steady and fluctuating) corresponding to most severe service and line test conditions.	-	A-4, A-6
irch.	Role of surfaces in adsorption and dissociation of ${\rm H_2}$.	Study of effects of surface oxide, sulfide, and other contaminant films on mechanisms of ${\rm H}_2$ entry into metals.	-	A-4, A-6
testing.	Stress corrosion cracking of steels in liquid NH ₃ .	Long-term delayed failure tests of new and used pipeline steels and welds in NH ₃ as a function of contaminants, structure and yield strength of steel, stress, notches, flaws, temperature and NH ₃ pressure.	Problem especially relevant to higher strength pipelines. Data produced also relevant to Items 2.3.1, 4.2.1 and 4.2.2	D-1.1
pipe erch.	Stress corrosion cracking of steels in liquid NH_3 .	Fuil-scale testing of new and used pipelines in pure and contaminated NH ₃ environments at temperatures and pressures corresponding to most severe service and line-test conditions.	-	D-1.1
perch.	Stress corrosion cracking of steels in liquid NH ₃ .	Studies to define the limits of the problem of the SCC of steels in liquid ${\rm NH}_3$.		D-1.1
testing.	Possible environmental effects of CO at high pressures for long times on pipeline materials.	Long-term tests of new and used pipe- line steels to establish if any deterioration of their mechanical properties can occur in pure or contaminated CO at high pressures.	_	E-1
	None.	None.	-	E-2
testing and	H ₂ environment effects over long times on higher strength pipeline steels.	Long-term testing of candidate pipe- line steels and welds in pure H ₂ and H ₂ contaminated with minor quantities of O ₂ , H ₂ O, odorants and illuminants. Test data required from -60°F to · 160°F on notched and unnotched tensile strength, fracture toughness, delayed failure under static load and low cycle fatigue.	Extension of programs under Items 1.1.1 to higher strength candidate pipeline steels.	A-5,1, A-5.3, A-6
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Table VIII-3 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE TRANSP

Item	Node of					
No.	Transportation	Fuel	Problem Area	Type of Solution	Materials Problem	Mate
2.	NEW PIPELINES (Cont.) (
2.1.2	H ₂ pipelines. (Continued)	H ₂ gas.	Protection of pipeline materials from H ₂ environments.	•	Fabrication and assembly of vented-lining type pipe.	Feasibili lining ty assembly.
2.1.3				Materials development, fabrication development.	Development and application of coatings with low ${\rm H_2}$ permeability.	Determine steels co etc. Inv mechanica after exp
2.1.4			Nonmetallic pipeline materials.	Materials selelction	Long-term mechanical behavior of plastics and composites in high pressure H ₂ , H ₂ O, and saline environments.	Investiga polymers ments aft pressure ments.
2.1.5				Fabrication develop- ment.	Costly, slow fabrication. Materials selection.	Develop 1 methods.
2.2.1	Cryogenic H ₂ pipe- lines.	F ₂ liquid.	Piping systems for cryogenic transmis- sion service.	Design and materials engineering.	Materials selection.	Waterials neering d
2.2.2			The "Energy Pipe."	Engineering and system design.	Materials availability and economics.	Supporting of techni
2.2.3				Vaterials research and development.	Improved superconducting alloys and composites.	Not speci
2.3.1	NH ₃ pipelines.	NH ₃ liquid.	Acceptability of new metallic pipeline materials for transmission and distribution of NH ₃ .	Materials research and testing.	Stress corrosion cracking of steels in liquid MI3.	Long-term candidate welds, as
2.4.1	O ₂ pipelines.	By-product O ₂ gas.	Safe, large-scale, economic pipeline transmission and distribution of O ₂ .	Materials qualification	Long-term compatibility of low-cost materials of construction with θ_2 at high pressures.	Determina pipeline pipe coni
2.4.2.				Materials research.	Lack of fundamental knowledge concerning effects of long-term exposure to high pressure O ₂ on mechanical behavior of metals.	Studies of mand contamechanics dition.
2.4.3				Engineering and system design; hazard analysis economic analysis	·	Supportir aspects of economic
2.5.1	Cryogenic O ₂ pipelines.	By-product O_2 liquid.	Safe distribution of liquid O_2 by pipeline.	Materials selection	Compatibility of metals and polymers with liquid \mathbf{U}_2 .	None.

Table VIII-3 (Continued)		
ND TESTING NEEDED TO SUPPORT T	THE THANSPORTATION OF NEW FUELS		
Nutariale Droblum	Matarials R D and T Youds	Pamark e	Report Reference Vol. 2, Sect. VI
With Training to the state of t	wateriars it, b. and r weeks	NCMAT NO	VOI. 2, Sect. VI
vented-lining type pipe.	reasionity study of low-cost vented- lining type pipe and methods of assembly.	-	A-5.1, A-5.3, A-6
Development and application of coatings with low H_2 permeability.	Determine rate of entry of $\rm H_2$ into steels coated with Cd, Pb, Sn, glasses, etc. Investigate deterioration of mechanical properties of coated samples after exposure to $\rm H_2$.	Exploratory study only.	A-5.1, A-5.3, A-f
Long-term mechanical behavior of plastics and composites in high pressure H_2 . H_2O , and saline environments.	Investigate mechanical properties of polymers and glass fiber reinforcements after long term exposure to high pressure H_2 , $\mathrm{H}_2\mathrm{O}$, and saline environments.	${\rm H_2}$ as internal pipe environment; ${\rm H_2O}$, saline as external pipeline environment.	A-5.2, A-5.3,
Costly, slow fabrication. Materials selection.	Develop low-cost, rapid fabrication methods.	-	A-5.2
Materials selection.	Materials testing to support engineering development.	Short-distance pipelines only. Some scale-up of state-of-the- art technology.	B-1
Materials availability and economics.	Supporting studies of materials aspects of technical and economic feasibility.	Concept development will require cooperation between groups concerned with hydrogen fuel and groups concerned with electric power distribution.	B-5, B-6
Improved Superconducting	Not specifically identified.	_	B-5
Stress corrosion cracking of steels in liquid NH ₃ .	Long-term delayed failure tests of candidate new pipeline steels and welds, as for Item 1.2.1.	See Item 1.2.1 Data also relevant to Items 4.2.1 and 4.2.2.	D-1.1
Long-term compatibility of low-cost materials of construction with \mathbf{O}_2 at high pressures.	Determination of ignition hazards for pipeline steels in actual or simulated pipe configurations.	Large-scale O ₂ pipeline system might be important aspect of overall hydrogen economy.	F.
Lack of fundamental knowledge concerning effects of long-term exposure to high pressure O ₂ on mechanical behavior of metals.	Studies of effects of long-term exposure of metals to high-pressure, pure and contaminated O ₂ on their mechanical behavior and surface condition.	-	F.
-	Supporting studies of materials aspects of engineering, hazard, and economic evaluations.	-	F.
Compatibility of metals and polymers with liquid \mathbf{O}_2 .	None.	State-of-the-art technology. Only short distance pipelines likely.	F.
VIII-19			
	Waterials Problem Fabrication and assembly of vented-lining typs pipe. Development and application of coatings with low H2 permeability. Long-term mechanical behavior of plastics and composites in high pressure H2. H2O, and saline environments. Costly, slow fabrication. Materials selection. Materials selection. Materials availability and economics. Stress corrosion cracking of steels in liquid NH3. Long-term compatibility of low-cost materials of construction with O2 at high pressures. Lack of fundamental knowledge concerning effects of long-term exposure to high pressure O2 on mechanical behavior of metals. Compatibility of metals and polymers with liquid O2.	Fabrication and assembly of vented-lining type pipe. Development and application of coatings with low H2 permeability. Development and application of coatings with low H2 permeability. Long-term mechanical behavior of plastics and composites in high pressure H2. H20, and saline environments. Costly, slow fabrication. Materials selection. Materials selection. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economics. Develop tow-cost, rapid fabrication methods. Materials availability and economic feasibility. Develop tow-cost, rapid fabrication methods. Materials availability and economic feasibility. Develop tow-cost, rapid fabrication methods. Supporting studies of materials aspects of engine feasibility. Develop tow-cost, rapid fabrication methods. Not specifically identified. Develop tow-cost, rapid fabrication methods. Not specifically identified. Develop tow-cost, rapid fabrication methods. Supporting studies of materials aspects of engine feasibility. Develop tow-cost, rapid fabrication methods. Studies of effects of long-term exposure to high pressure of metals to high-pressure, pure and contactions. Supporting studies of materials aspects of engine feasibility of metals and polymers with liquid Q2.	TESTING NEEDED TO SUPPORT THE THANSPORTATION OF NEW FUELS **Materials Problem** **Waterials Problem** **Waterials Problem** **Waterials Problem** **Waterials Problem** **Waterials Problem** **Particution and assembly of problems and assembly of coatings with low H ₀ assembly. **Development and application of coatings with low H ₀ steels coated with Cd, Pb, Sn, glasses etc. Investigate deterioration of assembly. **Long-term mechanical behavior of plastics and coaposites in high pressure H ₀ , H ₀ , and saline environments. **Long-term mechanical behavior of plastics and polymers and glass fiber reinforcements after caposure to H ₀ . Investigate mechanical properties of polymers and glass fiber reinforcements. **Betterials solection.** **Waterials solection.** **Waterials solection.** **Waterials solection.** **Waterials solection.** **Waterials solection.** **Waterials availability and economics.** **Supporting studies of materials sapetts Concept development will proper concerned with hydrogen fuel and groups concerned with hydrogen conditions. **Supporting studies of anterials and selds, as for Item 1,2.1. **Long-term compatibility of low-cost natorials of construction with O ₂ at high pressures. **Stress corrosion cracking of steels in liquid NH ₂ . **Long-term compatibility of low-cost natorials of construction with O ₂ at high pressures. **Long-term compatibility of low-cost natorials of construction with O ₂ at high pressure O ₂ on mechanical behavior of metals. **Studies of effects of long-term exposure to high pressures.** **Studies of effects of long-term exposure to high pressure O ₂ on mechanical behavior and surface condition. **Supporting studies of materials appet of overall hydrogen concerning effects of engineering, hazard, and conomic evaluations.** **Computability of metals and polymers with liquid O ₂ . **Studies of engineering, hazard, and conomic evaluations.** **Computability of metals and polymers with liquid O ₂ . **Studies of engineering, hazard, and

Table VIII-3 (Concluded)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE TRANSPOR

Item No.	Mode of Transportation	Fuel	Problem Area	Type of Solution	Materials Problem	Mate:
3.	PRESSURE VESSELS					
3.1.1	High pressure H ₂ vessels.	H ₂ gas.	Higher operating pressures.	Establishment of safe operating practices. Limitation of permissible H ₂ pressures.	Effects of H ₂ environment; on candidate pressure vessel materials.	Determine and low-cy candidate taminated pressure.
4.	SURFACE AND AIR TRANSPORTATION					
4.1.1	Barge, railroad car, tank truck.	H ₂ liquid.	H ₂ losses; cost.	Engineering design and development. Materials development.	Improved insulation.	None,
4.2.1	Various surface.	NH ₃ liquid.	Tank materials.	Materials research; materials selection.	Stress-corrosion cracking of quenched and tempered steels in liquid NH ₃ .	Materials mechanism
4.2.2		- · · · <u>- · · · · · · · · · · · · · · ·</u>				Testing of materials
4.3.1	Various surface.	Hydrazine.	Tank materials.	Materials selection.	Corrosion of container materials.	None.
4.3.2					Catalytic decomposition of N_2H_4 by container materials.	None.
4.4.1	Various surface.	Methanol.	None.		None.	None.
4.5.1	Ocean tanker.	H ₂ liquid.	Tanker design (liquid ${ m H_2}$ tanks not considered a problem area).	Engineering design.	None .	None.
4.6.1	Air tanker.	H ₂ liquid.	Aircraft design optimization.	Engineering design.	-	None.
4.6.2			L quid H ₂ tanks.	Materials selection and design optimization.	Righ strength, high stiff- ness, light weight, low thermal conductivity and low thermal expansion materials desirable.	Determine properties for candid at -423°F,
4.7.1	Transfer and delivery systems for use with surface and air tankers.	H ₂ liquid.	Heat leaks due to penetration of tank walls by transfer vent lines.	Engineering design; materials develop- ment.	Low thermal flux tubing.	Materials support fo programs.
4.8.1	Transfortation of hydrogen as metal hydrides	H ₂ gas.	Containment vessels.	Naterials selection.	H ₂ environment effects on container materials.	None.

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والماروة اوردة	Table VIII-3 (Conclud	ed)		
¢ .				
PMENT AND	TESTING NEEDED TO SUPPORT TH	IE TRANSPORTATION OF NEW FUELS		
				Report Reference
etion	Materials Problem	Materials R, D, and T Needs	Remarks	Vol. 2, Sect. VI
생 면 문				
	Effects of H ₂ environments	Determine threshold stress intensities	For small quantity distribu-	A-3, A-6
. ·	on candidate pressure	and low-cycle fatigue properties of	tion only.	
permis- ures.	vessel materials.	candidate materials in pure and con- taminated H ₂ gas as a function of gas		
		pressure.		
To Be Procedure to the Control of th				
eien and	Improved insulation.	None.	Improvements likely to be	B-2
steriuls	Improved insulations		provided by industry	-
E.			developments.	
Les.	Stress-corrosion cracking	Materials research to establish	See Vol. 1, Section III-D.1	D-1.2
ction.	of quenched and tempered steels in liquid NH ₃ .	mechanism and limiting conditions.	Overlaps Items 1.2.1	
A de la companya de l	stors in ridara unit.	Testing of candidate SCC resistant	Overlaps Items 1.2.1 and	
ige karjalisa ya karjalisa ka		materials and welds.	2.3.1	
ètion.	Corrosion of container	None.	Existing technology.	D-2
5°	materials.			
č N	Catalytic decomposition	None.	Existing technology,	D-2
Address and the CAC date and the CAC dat	of N ₂ H ₄ by container materials.			
£	None.	None.	Standard practice.	E-2
rign.	None.	None.	Low density cargo, Liquid H ₂	B-2
Š	-		tanks similar to LNG tanks.	
eign.	•	None.	Low density cargo.	
** **				n 0
etion imiza-	High strength, high stiff- ness, light weight, low	Determine physical and mechanical properties as needed for design	-	B-3
\$.	thermal conductivity and	for candidate composite materials		
	low thermal expansion	at -423°F.		
Š	materials desirable.			
	Low thermal flux tubing.	Materials development and testing support for engineering design	•	B-4
rop-		programs.		
Cop-				
2:				
etion.	H ₂ environment effects on container materials.	None.	Not attractive for H ₂ transportation on a	С
(p	LUNCOLUUM MELTILAID.			
Chemical I			large scale.	

TABLE VIII-4 MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE STORAGE OF NEW FUELS

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE :

No.	Type of Storage	Fuel	Problem Area	Type of Solution	Materials Problem	
1.	NATURAL FORMATIONS (Natural or modified)				
1.1.1	Depleted oil or gas reservoirs, aqui- fers, natural or mined caverns.	H ₂ , CO gases, and NH ₃ .	Porosity of formation.	Grouting.	None.	None.
1.2.1	Underwater storage.	H ₂ , CO gases.	Design and construc- tion.	Engineering design; materials selection.	Materials employed must be resistant to aqueous (or salt water) corrosion and effects of H ₂ (or CO) environments.	None
2.	MAN-MADE BULK STORAGE SYSTEMS (GASES)					
2.1.1	Pipelines, capped pipe, welded steel vessels for moder- ate pressures (≤ 2000 psi).	H ₂ gas.	Acceptability of low and medium strength steels for H ₂ storage.	Materials testing and selection.	${\rm H_2}$ environment effects over long times on low and medium strength steels.	Long- and w from requi unnot tough stati
2.1.2			Protection of materials of construction from H_2 environments.	•	Fabrication and assembly of vented-lining type vessels.	Develoand a
2.1.3				Materials development, fabrication develop- ment.	Development and application of coatings with low ${\rm H_2}$ permeability.	Detern steel: etc. mech u
2.2.1	Welded alloy vessel for high pressures (> 2000 psi).	H, gas.	Acceptability of high- strength alloys for H ₂ storage.	Materials testing and selection.	H ₂ environment effects over long times on high-strength alloys.	Determon med
2.3.1	Filament-wound fiber-reinforced plastic vessels.	H ₂ gas.	Acceptability of high- strength FRP vessels for H ₂ storage.	Materials testing.	Long-term mechanical behavior of FRP in high-pressure H_2 .	Determent to his
2.3.2				Fabrication develop- ment.	Present fabrication methods are expensive, slow, and size-limited.	Develo scale
2.4.1	Lined, fiber-rein- forced and/or polymer-impregnated	H ₂ gas.	Design and construction.	Engineering development and evaluation; mater- ials testing.	Long-term materials data for design is lacking.	Determenties compos

cement vessels.

	Table VIII-4			
NT .	AND TESTING NEEDED TO SUPPORT	THE STORAGE OF NEW FUELS		
				Report Referenc
	Materials Problem	Materials R, D, and T Needs	Remarks	Vol. 2, Section
	None.	None.	Practice similar to existing storage of natural gas.	B-1, C-2.1
<u> </u>	Materials employed must be resistant to aqueous (or salt water) corrosion and effects of H ₂ (or CO) environments.	None at this time.	Concept only.	B-i
d	${\rm H_2}$ snvironment effects over long times on low and medium strength steels.	Long-term testing of candidate steels, and welds in pure and contaminated $\rm H_2$ from -60°F to +160°F. Test data required includes notched and unnotched tensile strength, fracture toughness, delayed failure under static load, and low-cycle fatigue.	"Line-packing" of transmission pipelines is a standard method of natural gas storage. Data would permit specification of safe design and operating standards.	B-2, B-3
	Fabrication and assembly of vented-lining type vessels.	Development of low-cost fabrication and assembly methods.	Vented-lining techniques employed in high pressure, chemical process vessels containing ${\rm H}_2$.	B-2, B-3
it,	Development and application of coatings with low H ₂ permeability.	Determine rate of entry of $\rm H_2$ into steels coated with Cd, Pb, Sn, glasses, etc. Investigate deterioration of mechanical properties of coated samples after exposure to $\rm H_2$.	-	See Section VI-A 5.1
ıd		Determine effects of high-pressure ${\rm H_2}$ on mechanical properties of candidate high-strength alloys and welds.	-	B-2, B-3
	Long-term mechanical behavior of FRP in high-pressure \mathbf{H}_2 .	Determine effect of long-term exposure to high-pressure H ₂ on mechanical properties of FRP.	-	B-2, B-3
	Present fabrication methods are expensive, slow, and size-limited.	Develop low-cost, rapid, and large-scale fabrication methods.	-	B-2, B-3
ent	Long-term materials data for design is lacking.	Determine long-term mechanical properties of various types of cement composites.	Possible low-cost, large- scale storage vessels.	B-3
	VIII-21			

Table VIII-4 (Continued)

WATERIALS RESEARCH DEVELOPMENT AND TESTING NESDED TO SUPPORT THE STORA

Item	Type of Storage	Fuel	Problem Area	Type of Solution	Materials Problem	Mate
3.	BULK STORAGE SYSTEMS (LIQUIOS)	S				
3.1.1	Conventional systems (vacuum jacketed) up to 3×10^6 gal.		Tanks and auxiliary equipment.	Materials selection.	None.	None.
3.2.1	Metal or metal-lined concrete tanks > 3 x 10 ⁸ gal.	i H ₂ liquid.	Design and construction.	Engineering develop- ment; materials selection.	None.	None.
3.3.1	Reinforced plastic tanks.	H ₂ liquid.	Design and construction.	Engineering and fabrication development.	Present fabrication methods expensive, slow and size-limited.	Develop lescale fab
3.3.2				Materials testing.	Naterials data for design at cryogenic temperatures is lacking.	Determina ical prop ites at c
3.3.3				Materials development.	Chill-down stresses would be reduced by low-expansion materials.	Develop a composite trolled o advanced or zero
3.4.1	Fiber reinforced impregnated cement tanks.	H ₂ liquid.	Design and construction.	Engineering develop- ment and evaluation; materials testing.	Vaterials data for design at cryogenic temperatures is lacking.	Determina ical prop cements a
3.5.1	All liquid H ₂ bulk storage systems.	H ₂ liquid.	Insulation materials.	Engineering design and development; materials development.	Improved, low-cost internal and external insulation materials.	Materials engineeri
3.5.2			Auxiliary components.	Materials selection, component testing.	Effects of high-purity H ₂ environments on materials.	Test comp
3.6.1	Double wall, perlite insulated steel tanks.	NH ₃ liquid.	None using conventional low-strength steel construction.		None,	None.
3.6.2			Use of higher strength steels.	Materials research and testing.	Stress corrosion cracking of steels in liquid NH_3 .	Materials mechanism steels in
3.6.3						Testing o
3.7.1	Methanol bulk	Methanol	None.	_	Nore	•

storage tanks.

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100 m				
	Table VIII-4 (Continu	aed)		
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OSMENT	Table VIII-4 (Continu AND TISTING NEEDED TO SUPPORT Materials Problem	T THE STOPACE OF NEW FUELS		
				Report Reference
3on	Materials Problem	Materials R, D, and T Needs	Remarks	Vol. 2, Sect. VII
12 G				
don.	Vone,	None.	Existing technology.	C-1.1, C-1.5
don.				
lop-	None.	None.	Double or single wall	C-1.1, C-1.5
			construction.	
<u> </u>				
fabri-	Present fabrication methods	, , , , , , , , , , , , , , , , , , , ,	Strength requirements not as	C-1.1, C-1.5
abri- mt. Resident de la companya d	expensive, slow and size- limited.	scale fabrication methods.	great as Item 2.3.1.	
	Materials data for design	Determination of physical and mechan-	•	C-1.1, C-1.5
	at cryogenic temperatures	ical properties of candidate compos-		·
	is lacking.	ites at cryogenic temperatures.	For anallan Araba Ocaka	
ment.	Chill-down stresses would be reduced by low-expansion	Develop and test reinforced polymeric composites and structures with con-	For smaller tanks. Costs likely to be too high for	C-1.1
	materials.	trolled orientation of graphite or	large vessels.	
resident of the second		advanced organic fibers to give low or zero expansion coefficients.		
lop-				
	Materials data for design	Determination of physical and mechan-	-	C-1.1, C-1.5
tion;	at cryogenic temperatures is lacking.	ical properties of fiber reinforced cements at cryogenic temp atures.		
n and orials	Improved, low-cost internal and external insulation	Materials D and T support for engineering development.	•	C-1.4
	materials.	engineering development.		
ion,	Effects of high-purity H ₂	Test components in high-purity H2	-	C-1.1
	environments on materials.	at service conditions.		
ion,	None.	None .	Standard practice.	C-2.1
	-		· ·····	
	a.		0	
ch and	Stress corrosion cracking of steels in liquid NH ₃ .	Materials research to establish mechanism and limits of SCC of	See also Vol. 1 Sect. III- D.1 and Vol. 2, Sect. VI-	C-2.1
		steels in liquid NH3.	D.1. Same as 1tem 1.5.1	
		Markhau and and the comment	in Table VII(-1.	0.0.1
		Testing of candidate SCC resistant materials.	-	C-2.1
¥	None	-	Standard practice.	C-2.2
£	eringen Philippin (Schoolschauserings, erinisti (Schoolschausering, wied, de Sentre Schoolschausering			
	yrtt.oo			
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r E				

Table VIII-4 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE STO.

Item No.	Type of Storage	Fuel	Problem Area	Type of Solution	Materials Problem	Mate
4.	SMALL CONTAINER STORAGEGASES					
4.1.1	FRP vessels.	H ₂ gas.	Lighter vessels needed	Develop and qualify filament-wound rein- forced plastic tanks.	Long-term mechanical behavior of FhP in high pressure \mathbf{H}_2 .	Determing to high erties (
5.	SWALL CONTAINER STORAGELIQUIDS					
5,1.)	Vacuum-jacketed tanks.	H ₂ liquid.	Low-cost, mass-ploduced storage tanks.	Pasign and fabrication development.	None.	Material port for grams.
5.2.1	Insulated tanks.	NH ₃ liquid.	Low-cost, mass-pro- duced storage tanks.	Design and fabrication development.	None.	None.
5.3.1	Uninsulated tanks for hydrazine.	Hydrazine.	Very long-term (10 year) storage.	Design, materials development and testing, chemical studies.	Long-term compatibility of metallic and nonmetallic materials with hydrazine.	Developm inert ma
5.4.1	Uninsulated tanks for methanol.	Methanol,	None of significance.	-	None.	Nore.
6.	AIRCRAFT AND SPACE VEHICLE FUEL TANKS					
6.1.1	All aircraft and space vehicles.	H ₂ liquid.	Flight-weight tank design.	Engineering design and development using advanced composites.	Lack of design data for cryogenic temperatures.	Determine propertie at cryoge
6.1.2				Fabrication and materials develop-ment.	Chill-down stresses would be reduced by low-expansion materials.	Develop : composite trolled c advanced or zero e
6.1.3					Adhesive for use in contact with liquid H_2 .	Develop s
6.1.4					Flexible membrane materials for liquid H ₂ or H ₂ vapor barriers.	Develop h
6.2.1	Subsonic aircraft	H ₂ liquid,	Fuel tank insulation.	Engineering and materials development.	Safe, efficient insulation for temperature range ~423°F to ~180°F.	Develop a losed-po vermal c .ructura +180°F.

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2.5 2.7				
CH DEVELOPMENT		a.		
All facilities	Table VIII-4 (Continue	d)		
CH DEVELOPMENT	ANL FESTING NEEDED TO SUPPORT	THE STORACE OF NEW FUELS		
r-Solution	Materials Problem	Materials R, D, and T Needs	Remarks	Report Reference Vol. 2, Sect. Vol.
and qualify	Long-term mechanical	Determine effects of long-term exposure	•	B-2, B-3
wound rein- lastic tanks.	behavior of FRP in high pressure H ₂ .	to high pressure H ₂ on mechanical properties of FRP.		
2 -				
ad fabrication	None.	Materials development and testing sup-	Essenatial for use of liquid	C-1.2
ent.		port for engined ing development programs.	H ₂ by vehicles, small boats, portable equipment.	
	N		Par sawre cilarbuents.	
and fabrication	None.	None.	-	C-2.1
	Y 4	Development and Assistance & Markley	01	0.01
naterials ent and	Long-term compatibility of metallic and nonmetallic	Development and testing of highly inert materials or coatings.	Short-term small container storage of hydrazine poses	C-2.1
chemical	materials with hydrazine		no serious problems.	
	None.	None.	Corrusion may be a slight problem.	C-2.2
ing design and	Lack of design data for	Determine physical and mechanical	Storage times required are	C-1.3
ent using composites.	cryogenic temperatures.	properties of candidate composites at cryogenic temperatures.	relatively short.	
ion and	Chill-down stresses would	Develop and test reinforced polymeric	Samo as Item 3.3.3.	C-1.3
develop-	be reduced by low-expansion materials.	composites and structures with con- trolled orientation of graphite or		
	•	advanced organic fibers to give low		
	Adhesive for use in contact	or zero expansion coefficients. Develop adhesives for service at	-	C-1.3
	with liquid H_2 .	-423°F.		
	Flexible membrane materials for liquid H ₂ or H ₂ vapor	Develop high-strength flexible membrane materials for service at -423°F.	-	C-1.3
ion and develop-	barriers.			
ing and	Safe, efficient insulation	Develop and test fire-resistant	Foams may be fiber rein-	C-1.3, C-1.4
s development.		closed-pore foams with very low thermal conductivity and good	forced.	•
	.20 1 to +100 F.	structural strength from -423°F to		
		+180°F.		
h hases				
	VIII-23			
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Table VIII-4 (Continued)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE STO

Item	Tune of Stance	Post	Duchlan Anca	There are Call its an	Made and all a Double and	****
No.	Type of Storage	Fuel	Problem Area	Type of Solution	Materials Problem	Mate
6.	AIRCRAFT AND SPACE VEHICLE FUEL TANKS (Concluded)					
5.3.1	Supersonic sircraft.	H ₂ liquic.	Fuel tank insulation.	Engineering and materials development.	Hot-face insulation temperature may reach 350°F.	- Material. necering
3.4.1	Hypersonic aircraft.	H ₂ gas.	Fuel tank insulation.	Engineering and materials development.	Target for insulation operating range is -423°F to +650°F.	Materials neering c
7.	STORAGE OF NEW FUELS AS SOLIDS					
7.1.1	Storage of hydrogen us metal hydrides.	H ₂ gas	Technical and economic feasibility.	Cost/benefit analysis of hydride systems compared with liquid and gaseous H ₂ , for various storage capacitic; and applications.	Estimate materials component of system costs.	Assistanc technical study.
7.1.2			Lack of operational axperience.	Build, est, and develop engineering prototype systems based on Mg and FeTi,	Not applicable.	Assistanc
7.1.3			Performance improve- ments.	Materials research and development.	Absorption and desorption kinetics.	Studies c desorptic candidate
7.1.4						Investiga additions systems.
7.1.5					Limited cycle life, especially for deep discharges.	Examine ϵ H_2O , odor life.
7.1.6						Investiga charge re and bed c bed struc
8.	STORAGE OF BY-PRODUCT OXYGEN	r				
8.1.1		By-product O ₂ gas	Possible (gnition hazards.	Engineering studies.	None	None

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	Table VIII-4 (Continued) AND TESTING NEEDED TO SUPPORT			
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	Maka - 11779 4 (m			
A COLOR	Table VIII-4 (Continued)			
MENT	AND TESTING NEEDED TO SUPPOR	RT THE STORAGE OF NEW FUELS		
				Report Reference
	Materials Problem	Materials R, D, and T Needs	Remarks	Vol. 2, Sect. VII
mendings of the second of the				
K C	Hot-face insulation tempera-	Materials D and T support to engi-	-	C~1.3
nt.	ture may reach 350°F.	neeering development programs.		• • • • • • • • • • • • • • • • • • • •
	_			
ia M E	Target for insulation	Materials D and T support to engi-	Hot-face insulation tempera-	C-1.3, C-1.4
mat.	operating range is -423°F to +650°F.	neering development programs.	tures will depend on aircraft or vehicle design.	
Mis	Estimate materials compon-	Assistance in materials aspects of	Study needs to be done in	D-4
d	ent of system costs.	technical and economic feasibility	depth. Published comparisons	
td 2		study.	are not considered adequate.	
_				
Se de la companya de				
i E				
	Not applicable.	Assistance and consultation.	One program using FeTi	D-4
T 1.			initiated at Brookhaven National Laboratory.	
Ţ1.			•	
and	Absorption and desorption	Studies of H2 absorption, storage and	-	D-4
<u>ئىلىلە</u> دەخلىلە	kinetics.	desorption characteristics for new		
		candidate systems.		D 4
ا ماند الله الله الله الله الله الله الله الل		Investigate effects of minor alloy additions on performance of known	-	D-4
		systems.		
	Limited cycle life,	Examine effects of contaminants (O2,	-	D-4
	especially for deep	H ₂ O, odorants, illuminants) on cycle		
Ž.	discharges.	life.		
356		Investigate cycle life/depth of dis- charge relationships for various rates	-	D-4
į.		and bed conditions. Develop improved		
ţ. Į		bed structures.		
3				
	Yone	None	Analagous to current practice	E
į			with natural gas.	
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Table VIII-4 (Concluded)

MATERIALS RESEARCH DEVELOPMENT AND TESTING NEEDED TO SUPPORT THE STORAGE OF

Item No.	Type of Storage	Fue1	Problem Area	Type of Solution	Muterials Problem	Materials
8.	STORAGE OF BY-PRODUC GXYGEN	7	·			
8.2.1	Bulk storage as gas in man-hade systems.	By-product O ₂ gas.	Safety of storage vessels.	Materials qualifics- tion.	Long-term compatibility of low-cost materials of construction with O_2 at high pressures.	Determination materials of c ditions simula sure.
8.2,2				Materials research.	lack of fundamental knowledge concerning effects of long-term exposure to high pressure O ₂ on mechanical behavior of metals.	Studies of eff sure of metals and contaminat behavior and s
8.2.3				Engineering and system design; hazard analysis; economic analysis.	-	Supporting stu- aspects of eng and economic e
8.3.1	Bulk storage as liquid in ran-made systems.	By-product O. liquid.	Safety of storage vessels.	Materials selection.	Compatibility of metals and other materials with liquid \mathbf{O}_2 .	None.

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A STATE OF THE STA	Table VIII-4 (Concluded) AND TESTING NEEDED TO SUPPORT			
	Table VIII-4 (Concluded)			
WELOPMENT A	AND TESTING NEEDED TO SUPPORT	THE STORAGE OF NEW FUELS		
ntion	Materials Problem	Materials R, D, and T Needs	Remarks	Report Reference Vol. 2, Sect. VII
Diffica.	Long-term compatibility of low-cost materials of construction with \mathbf{O}_2 at high pressures.	Determination of ignition hazards for materials of construction under conditions simulating most severe exposure.	-	E
enerch.	Lack of fundamental knowledge concerning effects of long-term exposure to high pressure O_2 on mechanical behavior of metals.	Studies of effects of long-term exposure of metals to high pressure pure and contaminated G_2 on their mechanical behavior and surface condition.	-	Ε
ind system	-	Supporting studies of materials aspects of engineering, hazard, and economic evaluation.	-	E
ection.	Compatibility of metals and other materials with liquid \mathbf{O}_2 .	None.	Existing technology and practice. Safety standards must be enforced to prevent accidents.	E
arch.			•	
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IX RESEARCH RECOMMENDATIONS AND PRIORITIES

This section collates and reclassifies the materials research, development, and testing needs identified in Tables VIII-1 through VIII-4 into ten major programs as follows:

Table IX-1

Program A: Effects of Hydrogen on Materials

Table IX-2

Program B: Effects of Hydrogen Carriers and Partially Oxygenated Compounds on Materials

Table IX-3

Program C: Effects of Oxygen (By-Product) on Materials

Table IX-4

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Program D: High-Temperature Materials Studies

Table IX-5

Program E: Materials for Service at Cryogenic Temperatures

Table IX-6

Program F: Materials for Fuel Cells and Electrolyzers

Table IX-7

Program G: Materials for High Energy Density Batteries

Table IX-8

Program H: Catalysts (Excluding Electrode-Catalysts)

Table IX-9

Program J: Miscellaneous Materials Development and Fabrication

Table IX-10

Program K: Technoeconomic and Engineering Feasibility and Evaluation Studies.

A number of specific projects are described within each major program.

The relevance of each project to DoD is estimated and a priority is suggested. The tabular form employed and the significance of the individual table columns are described below.

Project Number (Column 1)

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Column 1 lists the project number which consists of the major program letter (A through K) followed by a number that identifies the individual project within the major program.

Project Description (Column 2)

Column 2 gives a project title and a short description of the suggested scope of the study. In some instances the project is divided into subprojects, identified by lower case letters in parentheses, e.g., (a).

Activity Type (Column 3)

Column 3 contains a simple description of the type of activity involved in the performance of the project; for example, "Basic research" or "Materials testing and interpretation."

Fuel (Column 4)

Column 4 lists the fuel or fuels to which the project relates.

Problem Area (Column 5)

Column 5 gives an abbreviated description of the problem area to which the proposed project or subproject relates. A horizontal line running from Column 4 (Fuels) through Column 10 separates the information contained in these columns according to the problem area indicated in Column 5. In some instances, a single project, for example, project A-1 in Table IX-1, is related to several problem areas. In some other cases, individual subprojects are related to separate problem areas; in this instance, the relationship is clarified by the subproject identifying letter given before the problem area description, for example, project A-5(a) and A-5(b) in Table IX-1.

Reference to Section VIII (Columns 6 and 7)

These columns list the table and item numbers of the materials research, development, and testing needs identified in Section VIII that the project or subproject described in Column 2 aims to solve. The relevant textual matter of Sections IV through VII in Volume 2 can be traced using these Section VIII item numbers and table numbers.

Relevance to DoD (Column 8)

This column indicates our judgment of the relevance of the project or subproject to DoD requirements. The relevance is indicated as "high," "moderate." or "low" and reflects the highest degree of relevance among the items listed in Column 7. (These items do, of course, vary in their relevance to DoD even though their solution is provided by the same project or subproject.)

Priority (Column 9)

Column 9 indicates our rating on a scale of 1 (high) to 5 (low) of the priority that should be accorded to the project or subproject in the context of the specific problem area shown in Column 5. A project may be assigned a high priority in relation to one problem area and a low priority in regard to another, since given problem areas will be of varying importance and may require solutions in different time-frames. The listed priority rating is based on a combination of the importance of the problem, the urgency of the need for a solution and the relevance of the problem to the DoD. It must be emphasized that the priority judgments represent our interpretation of the information uncovered during the performance of this study and that new information or changes in DoD policy might necessitate a corresponding update of the priority ratings.

Remarks (Column 10)

Explanatory comments are made in this column either to clarify the relevance or priority ratings in Columns 8 and 9, or to provide additional general information on the problem area or the nature of the project.

IX-A. DISCUSSION OF RESEARCH RECOMMENDATIONS

The suggested materials research, development, and testing projects are briefly described in the following paragraphs under the programs of which they are a part. An overall summary (Section IX-B) discusses the most important materials aspects of the use of new fuels in advanced energy systems.

1. PROGRAM A: EFFECTS OF HYDROGEN ON MATERIALS

In spite of the extensive literature on the effects of hydrogen on metallic materials, this study has shown that there are still serious deficiencies in our knowledge, particularly with regard to the behavior of materials of practical engineering significance under conditions that might be encountered in a hydrogen economy. This is especially true of the use of aerospace materials in hydrogen-fueled aircraft or rockets and of the use of general engineering materials for pipeline transmission and distribution of hydrogen. While the materials requirements for these two application areas are completely different the two cases have in common the fact that the materials employed are used at the highest possible level of performance. This common performance requirement is dictated in the case of aerospace materials by the need to minimize weight in flight structures of all kinds, and, in the case of pipeline materials, is a consequence of economic considerations.

Projects A-1 and A-2 relate to the use of aerospace structural materials in hydrogen environments. These are the materials employed in components extending from on-board fuel tanks through the point where the hydrogen fuel is burned. The materials testing and interpretation work recommended would encompass temperatures ranging from liquid hydrogen temperatures up to 1500°F and to significantly higher

temperatures in special cases, such as those that might occur if hydrogen were used to cool the hot components of turbine or rocket engines. Because of the different testing techniques required, the measurement and interpretation of the mechanical properties of aircraft structural materials in hydrogen environments has been divided into two separate projects, projects A-1 and A-2. Project A-1 is related to property measurements from -423°F to 300°F, while project A-2 is related to temperatures from 300°F upwards. For both projects, tests should be conducted both in an inert atmosphere, such as helium, and in hydrogen, to obtain an indication of the property degradation due to the hydrogen environment. Much of this work would extend and continue such programs as those summarized in Table IV-2 (Volume 2). Since it is likely that advanced composite materials with polymeric as well as metallic matrices may be employed over the low and moderately elevated temperature ranges, we believe testing of the e materials should be included, since no definitive information appears to exist as to the behavior of this category of materials in hydrogen environments.

The information obtained from the suggested test projects can be employed to determine the critical stresses (or stress intensities) below which the materials can be used safely when exposed to the most severe environmental conditions likely to be encountered in service. However, these design data will be of more value if the mechanisms causing property degradation are fully understood. For this reason, an important part of both projects will be to analyze and interpret the test results in terms of the composition, microstructure, and surface condition of the material, and the details of the test environment.

The effects of hydrogen environments on pipeline materials is the highest priority item under Project A-3: Hydrogen Environment Effects on Engineering Materials. The enormous capital investment in pipeline transmission and distribution systems makes it essential that a conclusive answer be provided to the question of whether or not present-day and future pipeline steels (and welds) are acceptable for use with hydrogen, and if so, what factors of safety should be applied to the design of the system.

Three aspects of the effects of hydrogen on the behavior of aerospace and general engineering materials are worthy of special mention:

• Hydrogen-assisted fatigue cracking.

- The effect of contaminants in the gas on the mechanical behavior of materials in hydrogen environments.
- The role of surface films and surface contaminants on the adsorption and dissociation of hydrogen.

The first two topics have been included in the mechanical properties investigations covered by Projects A-1, A-2, and A-3. The third topic is the subject of a separate project (A-6). The results of this project will have an important bearing on understanding the mechanical behavior of practical materials in hydrogen environments under actual service conditions.

Permeation of hydrogen through metallic and nonmetallic materials is not of general importance; however, the two specific studies shown under A-5 have been included. Project A-5(a) relates to the particular requirements involved in the use of hydrogen as the working fluid for Stirling cycle engines; its importance will depend on the extent of industry and DoD interest in this type of engine.

Project A-5(b) relates to the possible use of protective coatings to prevent the entry of hydrogen into metals. This project should be exploratory in nature, since the possibilities of practical success for such an approach would appear to be limited.

While most interactions of hydrogen with metals are considered as having unfavorable implications, the use of hydrides for storage of hydrogen represents an attractive possibility for utility peak shaving requirements and possibly for the on-board storage of hydrogen fuel for vehicles. Continuation and extension of metal hydride studies under Project A-7 has been assigned a priority rating of 2, but the level of effort should, in our view, depend on the outcome of technoeconomic and prototype studies to determine the potential advantages of this system compared with gaseous or liquid fuel storage.

In addition to the extensive materials testing studies identified under Project A-?, new or existing pipelines cannot be recommended for use with hydrogen until extensive full-scale pipe testing has been safely completed. This activity is identified under Project A-8 and is accorded a priority equal to the associated materials testing studies, as they relate to pipeline materials.

Project A-9 includes various materials development, testing, and consultation activities required to support engineering development and component testing programs in a variety of different problem areas. Since the potential effects of hydrogen on a wide variety of materials may be deleterious to varying degrees it is essential that adequately qualified materials experts are involved at all stages of the engineering design, development, testing, and production of equipment or components exposed to hydrogen environments to ensure maximum reliability and safety of operation.

It is our overall conclusion that the effects of hydrogen on materials will not constitute an insurmountable barrier to the safe and effective use of hydrogen as a fuel for military, industrial, commercial and residential use. However, before hydrogen fuel can be successfully introduced, extensive additional information must be developed. Program A represents a tentative suggestion of the materials research, development, and testing programs required to provide this information.

2. PROGRAM B: EFFECTS OF HYDROGEN CARRIERS AND PARTIALLY OXYGENATED COMPOUNDS ON MATERIALS

The use of the various hydrogen-derived fuels constitutes an alternative to the direct use of hydrogen itself. These alternative fuels are grouped as hydrogen carriers and partially oxygenated compounds. The principal hydrogen carrier is armonia, with hydrazine second in importance; the hydrogen carriers borane and silane also discussed in Volumes 1 and 2 of this report are not considered of sufficient importance to justify significant research effort at this time. The partially oxygenated compounds carbon monoxide and methanol derived from the hydrogen reduction of nonfossil carbon dioxide are also considered as alternatives to hydrogen fuels. Program B, Table IX-2 summarizes research, development, and testing programs considered necessary to support the use of these alternative fuels.

The choice of ammonia as an alternative fuel poses few serious materials problems, with one important exception: the stress corrosion cracking of steels in liquid ammonia, which now constitutes an unsolved problem important to the transportation of ammonia by tank truck.

The basic research proposed under Project B-1(a) and the materials testing and interpretation investigations proposed under Project B-1(b) are considered of high priority with regard to the transportation of ammonia by pipeline or surface methods. At the present time the actual extent of the phenomenon is not known and investigations to determine the limits of stress corrosion cracking with respect to the steel composition and microstructure, stress level and type of stressing. and the type and concentration of contaminants deliberately or accidentally present in the liquid ammonia should be instituted at an early date. These basic studies should be accompanied by long-term delayed-failure and fatigue tests of presently used and future candidate materials for ammonia pipeline or transportation vessels to select and qualify suitable materials of construction and to define safe materials design criteria. Full-scale testing of pipeline sections would also be necessary before the safety of new or existing pipelines for ammonia transportation could be assured.

Hydrazine is not, in our view, likely to become a major fuel for general use. For most military uses of hydrazine, adequate materials information already exists but additional compatibility studies are required for applications involving the long-time storage of hydrazine for periods of up to ten years. These investigations are included in Project B-3.

Of the partially oxygenated fuels, methanol does not appear to have any deleterious effects of importance on materials, with the exception of the well-known stress corresion cracking of titanium in methanol environments. The reactions of carbon monoxide with metals at elevated temperatures are well understood and common materials of construction are generally considered to be inert to carbor monoxide at

near-ambient temperatures and pressures. However, some uncertainty exists regarding the long-term effects of high-pressure carbon monoxide on the mechanical properties of metals. It is considered advisable to institute a materials testing and interpretation project to establish whether any deterioration of the mechanical properties of metals occurs under these conditions in pure and contaminated carbon monoxide. The priority accorded to this program at the present time is low, although it may be advisable to raise the priority if the general use of carbon monoxide fuel is projected.

Thus, the most important materials research and development projects concerned with the effects of hydrogen carriers and partially oxygenated compounds on materials are those directed towards solving the problem of the stress corrosion cracking of steels in liquid ammonia. We believe that significant support should be accorded these projects at an early date.

3. PROGRAM C: EFFECTS OF OXYGEN (BY-PRODUCT) ON MATERIALS

The mechanical properties of engineering materials are normally measured in air and it is assumed that oxygen environments do not change the property values. However, this assumption has not been adequately substantiated for long-term exposure to high-pressure oxygen when flaws or cracks are present in the material or when it is subjected to high-or low-cycle fatigue. A fundamental project to provide definitive information on this point is therefore suggested as a necessary preliminary to the large-scale use of oxygen and is relevant to the safet; of existing systems. Safety considerations have also led us to recommend a study of ignition hazards in oxygen pipelines containing surface cracks or flaws when subjected to rapid stressing, impact, or fatigue. (Project C-2.)

Probably the most important topic for study in connection with the large-scale general distribution and use of by-product oxygen is the question of whether such distribution is, in fact, technically and economically feasible. To arrive at an early judgment of this question, an in-depth engineering system study including hazard analysis and economic analysis is considered desirable.

4. PROGRAM D: HIGH-TEMPERATURE MATERIALS STUDIES

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Once the hydrogen fuel reaches the point of combustion, we have assumed that the effects of hydrogen itself on materials are no longer of concern. In some cases, momentary exposure of materials to hydrogen-containing combustion gases may exist where the mixture is momentarily fuel-rich but we consider that such transient situations can generally be neglected.

and of fossil fuels are the somewhat higher flame temperatures experienced with hydrogen, the different combustion conditions, and the fact that the combustion gases have a higher water content—pure water in the case of hydrogen combustion with oxygen. This last factor is approached from two different viewpoints in projects D-1 and D-2. Project D-1 proposes a fundamental investigation of the kinetics of reaction of candidate gas turbine materials with high water content environments. Silicon nitride and silicon carbide are considered important materials to be examined in this study because of the considerable government support of programs aimed at the use of these ceramics in both large and small gas turbines. In addition to this basic research project we consider it necessary to institute an extensive materials testing and evaluation project to determine the high-temperature mechanical and environment endurance of present and future candidate gas turbine materials in actual

or simulated hydrogen/air and hydrogen/oxygen combustion gases. Both of these projects will be important to advanced gas turbine design and development and are therefore considered of high relevance to the Department of Defense and have been accorded a 1 priority. Project D-2 consists essentially of an expansion of existing gas turbine materials programs to include the modified environment due to the change of fuel from hydrocarbon to hydrogen.

If water cooling of high-temperature gas turbine components, or components of oxygen/hydrogen combustion systems, is found to be feasible, the cooling channels will be subjected to high-velocity, high-temperature water which could produce rapid oxidation or corrosion due to the accelerating effect of erosion. Evaluation of this possibility is recommended if water cooling is adopted for high-temperature components.

5. PROGRAM E: MATERIALS FOR SERVICE AT CRYOGENIC TEMPERATURES

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Use of liquid hydrogen fuel will involve the extensive use of materials and components at temperatures near -423°F. While considerable data exists on the low-temperature behavior of a wide variety of materials, much of the information does not extend to or below the boiling point of hydrogen. An extensive project of materials property determination and evaluation will be required to support engineering design and development of equipment associated with the use, production, transportation, and storage of liquid hydrogen. These investigations grouped in Project E-1, will be of moderate to high relevance to DoD and of variable priority according to the specific application. Work related to aircraft liquid hydrogen fuel tanks will be of special concern to DoD and of high priority.

Materials development is recommended in two separate areas in connection with aircraft and space vehicle liquid hydrogen fuer tanks. In the first area (Project E-2), possibilities exist of reducing chill-down stresses in tanks and associated structures by the development of polymeric composite materials with low thermal conductivity and low or zero thermal expansion coefficients. Low thermal expansion coefficients can be obtained by the use of selected and accurately controlled orientations of reinforcing fibers of graphite or advanced organic fibers such as Kevlar 49^R that have negative coefficients of expansion. While these materials are likely to be expensive, the potential advantages of such materials and structures probably justify significant development work of high priority.

Highly efficient cryogenic insulation systems are of obvious importance in the handling of liquid hydrogen. While significant advances have been made in recent years, there appear to be two areas that would justify further materials development effort. The first area is the development of improved low-cost insulation systems for bulk liquid hydrogen storage (Project E-3). For the second area, insulation systems for aircraft and space vehicle liquid hydrogen tanks, continuation and expansion of existing projects is recommended. Two special requirements exist in this case. The first is for a high degree of fire resistance, and the second, applicable to supersonic and hypersonic aircraft, is that the insulation be able to function with high hot-face temperatures, which may in some instances reach 650°F. Materials support for the engineering development of improved insulation systems for this application is considered of high relevance to DoD and has been assigned a high priority.

6. PROGRAM F: MATERIALS FOR FUEL CELLS AND ELECTROLYZERS

To allow fuel cells to realize their potential for wide application in a nonfossil fuel economy, improved materials will be required for electrocatalysts, electrode structures, and electrolyte matrices. At the present time, it appears that alkaline hydrogen fuel cells (including regenerative types) are the most attractive class of cells for use in a hydrogen-based fuel system and high priorities in the materials research and development have therefore been accorded to programs associated with fuel cells of this type.

Advanced electrolyzers based on fuel-cell technology appear at the present time to have the best potential for reducing the basic cost of hydrogen fuel. In our view, the status of development of thermochemical splitting methods for the production of hydrogen is not yet sufficiently advanced to permit realistic estimates of the cost of hydrogen produced by this method. Since the feasibility and acceptability of a nonfossil hydrogen economy will depend to a large extent on the cost of hydrogen in comparison with the cost of alternative fuels, we believe that the materials-related programs in support of advanced electrolyzer development may constitute the most important scientific and technical effort related to the hydrogen economy.

Because of the close ties between advanced electrolyzer and advanced fuel-cell technology, these two subjects have been included in a common program. Projects F-1 through F-4 are concerned with improvements in electrocatalysts, a critical area for improving the efficiency of both fuel cells and electrolyzers. Projects F-1 and F-2 are principally directed at reducing the capital cost of the fuel cell or electrolyzer systems by lowering the loading of noble metal

catalysts required or by developing nonnoble metal catalysts of high activity and durability. Special purpose electrocatalysts included in Project F-4 are generally accorded lower priorities. Project F-3 is concerned with work of general application of both a fundamental and empirical nature. Fundamental studies of electrocatalysts are required to provide a scientific basis for electrocatalyst development. In this connection it should be pointed out that electrocatalyst behavior and the mechanisms of the electrocatalytic process differ markedly from those for general chemical process catalysis.

Problems associated with electrode materials and electrode structures and with the matrix or diaphragm materials are the subject of the investigations listed in Projects F-5 and F-6. The highest priority in these two projects is accorded to efforts to increase the operating temperature of alkaline fuel cells and alkaline electrolyzer systems, since in both cases higher temperature operation would result in increased efficiency.

Project F-7 involves additional basic materials research and materials development studies associated with the ionic behavior of solids. Project F-7(a) is directed at uncovering inorganic solids that have good ionic conducting properties at moderate temperatures. A significant breakthrough in this area could result in a reduced operating temperature for inorganic solid electrolyte fuel cells and electrolyzers, and by so doing, increase their efficiency and reduce the general materials problems of high-temperature operation. Research concerned with ion-conducting behavior in solid polymeric materials could make important contributions to increasing the efficiency and reducing the cost of the attractive solid polymer electrolyte electrolyzer systems. Among the general materials problems associated with the construction of fuel cells and electrolyzers, the most important is concerned with

the frame materials for alkaline electrolyzers. The presently used material, polysulphone is limited to 150°C operation. If higher temperature operation is to be obtained, improvements in the frame materials must accompany other advances in the technology.

The materials research and development associated with fuel cells and electrolyzers generally is of such a nature that it must be conducted in close collaboration with the electrochemical and engineering research and development activities. We suggest this consideration should be borne in mind in the planning and funding of materials research, development, and testing programs associated with fuel cell and electrolyzer development.

7. PROGRAM G: MATERIALS FOR HIGH ENERGY P. SITY BATTERIES

High energy density batteries are generally a topic of high or moderate relevance to the Department of Defense since portable power supplies are an essential requirement in almost all defense systems. Our review of this area clearly indicates that materials problems encountered in the development of a battery system are highly specific to the particular combination of electrodes and electrolyte involved in that system. Solutions developed to materials problems for one battery system are unlikely to apply to other systems because of the different environmental compatibility requirements. As a consequence, we believe that the materials research and development projects included under Program G should be undertaken as an integral part of battery development programs rather than as independent, materials-oriented studies. A possible exception to this generalization is Project G-2, in which studies of the ion-conducting behavior of solids are proposed. This work bears a close relationship to similar topics proposed in

Program F (Materials for Fuel Cells and Electrolyzers) and some combination of the research effort in these topics would be possible and advantageous.

8. PROGRAM H: CATALYSTS (EXCLUDING ELECTROCATALYSTS)

Three projects have been identified in the area of catalysis. The first of these relates to a specific DoD requirement for very long life catalysts for the catalytic decomposition of hydrazine within small monopropellant rocket engines.

The second group of catalyst research and development studies is associated with the catalytic combustion of new fuels (Project H-2). Catalytic combustion offers a highly efficient low-temperature heat source for space and water heating. Moderate priority is accorded to work associated with fundamental and developmental studies of the catalytic combustion of hydrogen, while relatively low priorities are given to the catalytic combustion of ammonia or other alternative new fuels.

The third project in this program is related to the possible use of methanol or carbon monoxide as alternative new fuels. In either case, the production of carbon monoxide by the hydrogen reduction of nonfossil carbon dioxide is necessary and is accomplished by the reversed shift reaction. Although this reaction is well known, it has not been of commercial significance as has the forward shift reaction, which results in the formation of hydrogen from carbon monoxide and water. If carbon monoxide and/or methanol become adopted as general purpose fuels, the development of catalysts to improve process efficiencies will be highly desirable. A low priority is accorded to this project in relation to the production of carbon monoxide (and not methanol) as a general fuel, while a higher priority is associated with the production of methanol, which we regard as a more likely fuel, particularly for vehicle use.

9. PROGRAM J: MISCELLANEOUS MATERIALS DEVELOPMENT AND FABRICATION

In this program we have collected various materials development and fabrication projects not covered (or not covered completely) in other programs. がいたが、これでは、これでは、これでは、これでは、これできないできない。これできないできない。これでは、これでは、これでは、これでは、これできない。これできない。

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Because of its important relation to advanced gas turbines systems and rocket engine development, Project J-1, consisting of materials support for the development of advanced cooling systems, is accorded a high priority and is considered highly relevant to DoD.

Project J-2, the development of fiber-reinforced cement and concretes, is a project of wide general applicability. These materials may offer a significant alternative to large-scale steel construction for pressure vessels and storage tanks.

Vented-lining construction of high-pressure process vessels for hydrogen service has been adopted in the process industry to provide protection for the steel pressure shell by the use of a thin hydrogen-resistant lining material. This type of construction is too expensive for general use but it is possible that low-cost manufacturing and assembly methods for such items as line pipe and plate for storage tanks might be feasible. Project J-3 is therefore suggested as an exploratory manufacturing development study program that could make possible the use of highly efficient high-strength steel structures in hydrogen environments.

Project J-4 includes three deve opment studies related to the use of fiber-reinforced composites for a variety of large and small pipe and tankage applications. Project J-4(c), also listed

in Program E, is accorded the highest priority in view of its relevance to DoD in relation to aircraft and space vehicle liquid hydrogen fuel tanks. Project J-5 is a small project of significance with regard to moving components operating in hydrogen environments and is particularly related to hydrogen compressors and hydrogen pumps. The relatively low priority accorded this project could be increased if engineering experience indicates a higher level of need.

10. PROGRAM K: TECHNOECONOMIC AND ENGINEERING FEASIBILITY AND EVALUATION STUDIES

In the performance of this study, we have uncovered three areas in which we feel necessary policy or strategy judgments require extensive technoeconomic or engineering feasibility and evaluation studies and analyses. The first of these, accorded the highest priority, relates to the production of hydrogen by electrolytic methods and to the large-scale use of fuel cells. The large-scale manufacture of these types of equipment will introduce new requirements for specialty polymers and ceramics, and noble and nonnoble electrocatalysts. Of particular concern is the possible need for vast quantities of platinum, far in excess of current usage or production. Questions of materials availability are likely to be critical in policy decisions relating to competing fuel cell and electrolyzer technologies and to general defense materials requirements. Project K-l is therefore considered highly relevant to DoD and has been accorded a priority of 1.

The second project suggested relates to the use of by-product oxygen. A critical element in judgments as to the desirability of distributing and using the oxygen produced as a by-product of hydrogen production processes as compared with venting it and burning the hydrogen

mission and distribution of by-product oxygen. Project K-2 is therefore regarded as a necessary preliminary step before significant funding is committed to development or engineering expenditures in this area. Project K-3 is more specific in nature and proposes a cost/benefit analysis of hydride systems compared with liquid and gaseous hydrogen storage for various applications, including peak shaving and vehicle uses. It is considered that adequate in-depth comparisons have not been made in this area and are needed before a definitive judgment as to the economic viability of hydride storage systems can be made.

IX-B. MATERIALS REQUIREMENTS FOR ADVANCED ENERGY SYSTEMS--NEW FUELS; SUMMARY AND OVERVIEW

This study sought to identify materials-critical aspects of the use, production, transportation, and storage of new fuels derived from nonfossil sources. Hydrogen was the principal new fuel studied; hydrogen-derived fuels considered were ammonia, hydrazine, boranes, silanes, carbon monoxide, and methyl alcohol. The materials implications of the use, transportation, and storage of oxygen (produced as a by-product in hydrogen generation) and of the use of active metals in batteries were also examined during the study. Previous volumes of this report are concerned with:

Volume 1: Interactions of Materials with New Fuels

Volume 2: Materials Aspects of the Use, Production, Transportation, and Storage of New Fuels

In this volume (Volume 3), the results of the study have been correlated and analyzed in Section VIII, while Section IX provides a listing of ten major research development and testing program areas within each of which specific projects are described. The relevance of these projects to DoD is noted and a judgment is given of their relative priorities in the context of the problem area to which they are related.

Of the four program areas--use, production, transportation, and storage--the materials requirements related to hydrogen production are probably the most important, since the viability of a hydrogen fuel economy depends above all on our ability to produce hydrogen with the most efficient use of energy and at the lowest possible cost. At the present time, the production of hydrogen by the thermochemical splitting of water is no. 12 a sufficiently advanced stage to permit any

clear assessment of its competitive status. It is therefore considered that the electrolysis of water is the most likely route by which hydrogen can be produced in the quantities required, and it is in this area that we believe a major research and development effort should be concentrated. The efficiency of electrolyzer systems is highly dependent on advances in electrocatalyst materials, materials for electrode structures and electrolyte matrices, and electrolyte materials.

The second question of key importance to the implementation of a hydrogen economy is whether hydrogen can be transported safely and economically in pipelines constructed of low-cost, readily available materials. This judgment will depend on defiritive technical information that does not now exist. An extensive program of materials testing and research should therefore be implemented at an early date on a scale sufficient to ensure that the necessary information is available when needed.

The use of hydrogen as a fuel in a wide variety of equipment does not appear to pose any insurmountable obstacles, although extensive materials research, development, and testing programs will be required to ensure maximum safety, reliability, and efficiency in hydrogen-using equipment. It is in the area of use that materials projects of highest relevance to DoD are found. Problem areas of particular importance to DoD requiring materials support include the use of hydrogen as an aircraft fuel, the further development of fuel cells for the direct conversion of hydrogen fuel to electrical energy, and the use of active metals in high energy density batteries.

Materials problems related to the storage of hydrogen and of the other new tuels do not appear to be a pacing factor. In this area, items of highest relevance to DoD are concerned with the on-board storage of liquid-hydrogen fuels in aircraft.

Among the alternative fuels, ammonia poses some problems in the area of transportation and storage that require materials research and testing. The economic production of methanol from nonfossil sources may present difficulties but this fuel appears otherwise attractive, particularly for vehicle use and presents no major materials problems.

In conclusion, we are confident that the materials requirements for advanced energy systems based on new fuels can be satisfied by a program of materials research, development, and testing of the type outlined in tabular form in this section of the report, coupled with the diligent and careful use of existing materials information.

 $\begin{tabular}{ll} Table & $\mathsf{IX-L}$ \\ \\ \mathsf{PROGRAM} & \mathsf{A:} & \mathsf{EFFECTS} & \mathsf{OF} & \mathsf{H_2} & \mathsf{ON} & \mathsf{MATERIALG} \\ \end{tabular}$

				Relates to:	Refere Section		
Project	Project Description	Activity Type	Fuel	Problem Area	Table	Item No.	Rele
A-1	Degradation of Vechanical Properties of Aerospace Structural Vaterials in Hydrogen Environments; -423°F to 300°F Properties: tensile, notched tensile, high- and low-cycle fatigue, fracture toughness, crack growth, delayed failure. Materials: Al-alloys, stainless steels, nickel- and cobalt-base alloys, brazing alloys, Tialloys, advanced composites, polymers, reinforced plastics. Variables: temperature, H ₂ pressure, contaminants, exposure time, strain rate, stress level, stress	Materials testing and interpretation.	H H ₂	Fuel supply and heat exchanger/gasifier for all types of H ₂ -burning gas turbines, hypersonic aircraft and rocket engines. H ₂ expansion turbines for aircraft and rocket fuel pumps. H ₂ -cooled turbine components. Stirling Cycle	1	1.2.1 1.3.1 1.4.1 2.1.1	Hig Mod Mod
	intensity, material composition and microstructure, welds.		H ₂	engines. Compressor components. Pressure vessels.	3	3.1.1	Low
			H ₂	Aircraft and space vehicle fuel tanks, air tanker.	3	2.3.1 4.6.2 6.1.1	Hig
A-2	Degradation of Mechanical Properties of Aerospace Structural Materials in Hydrogen Environments; 300°F to 1500°F With Extension to Higher Temperatures if Required. Properties: tensile, notched tensile, high- and low-cycle fatigue, fracture toughness (where	Noterials testing and interpretation.	H ₂	Fuel supply and heat exchanger/gasifier for all types of H ₂ - burning gas turbines, hypersonic aircraft, and rocket engines.	1	1.2.1 J.2.6 1.3.1 1.4.1 2 1.1 3.1.1	Rig
	applicable) crack growth, delayed failure, creep. Waterials: Al-alloys, stainless steels, nickel-		H ₂	H ₂ expansion turbines for aircraft and rocket fuel pumps.	1	1.3.1 3.1.1	Hig
	and cobalt-base alloys, Ch-alloys, brazing alloys, Ti-alloys, Si ₃ N ₄ , SiC, advanced composites, polymers. Variables: temperature, H ₂ pressure, contaminants, exposure time, strain rate, stress level, stress intensity, material composition and microstructure, welds.		H ₂	Stirling cycle engines.	1	6.3.1	Viod

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and the same		DRACDAN A. DEDRAMO OF	u ov :	MATERIALS			
		PROGRAM A: EFFECTS OF	n ₂ ON !	MATERIALS			
- Ecosts		Relates to:	Referen				
		normico to.	Table	7411	Relevance		
PRA.	Fuel	Problem Area	No.	Item No.	to DoD	Priority	Remarks
18 Per	н	Fuel supply and heat	1	1.2.1	High	1, 2	Continuation and extension of existing programs. See Vol. 1
¥.	H ₂	exchanger/gasifier	•	1.3.1		1, 2	Section III for review of current status, and Vol. 2, Tables
(2003)		for all types of H ₂ burning gas turbines,		1.4.1 2.1.1			IV-2, IV-3. Priority 1 for hypersonic aircraft, space vehicles; priority 2 for H ₂ -burning supersonic and
# AU		hypersonic aircraft					subsonic arreraft.
25.50	_ <u>u</u>	and rocket engines. H ₂ expansion turbines	1	1.3.1	High	1, 3	Necessary data for safe engine operation. Priority 1 for
	Н ₂	for aircraft and	•	1.3.1	17 14 14	1, 3	rocket engine fuel pump drive turbines. Priority 3 for H ₂
20.60		rocket fuel pumps.	1	1.2.6	Moderate	5	expansion aircraft engines.
No.	112	H ₂ -cooled turbine components.	1	1.2.6	Moderate	ง	Long-term engine development.
E. Service	${\rm H_2}$	Stiring Cycle	1	6.3.1	Noderate	Uncertain	
F22.73	H ₂	engines. Compressor com-	2	4.1.1	Low	5	industry interest. Likely to be industry activity, but may become of higher DoD
		ponents.					relevance and priority if energy depot concepts pursued.
	H ₂	Pressure vessels.	3 -1	3.1.1 2.2.1	liigh	1	Needed to establish safety standards.
				2.3.1			
	Н ₂	Aircraft and space vehicle fuel tanks,	3 -1	4.6.2 6.1.1	High	∶, 2	Priority 1 for hypersonic aircraft, space vehicles; priority 2 for H ₂ -burning supersonic and subsonic aircraft
		air tanker.					engines.
			_				
	H ₂	Fuel supply and heat exchanger/gasifier	1	1.2.1	High	1, 2	Priority 1 for hypersonic aircraft, space vehicles; priority 2 for H ₂ -burning supersonic and subsonic aircraft
1	Ì	for all types of H ₂ -		1.3.1			engines.
		burning gas turbines, hypersonic aircraft.		1.4.1 2.1.1			
Bace		and rocket engines.		3.1.1	!		
No. of the	H ₂	H ₂ expansion turbines for aircraft and	1	1.3.1 3.1.1	High	1, 3	Necessary data for safe engine operation. Priority 1 for rocket engine fuel pump drive turbines, priority 3 for H ₂
		rocket fuel pumps.					expansion aircraft engines.
्रमाप्तकः	H ₂	Stirling cycle engines.	1	6.3.1	Moderace	Unvertain	Relevance and priority depend on DoD vehicle and small engire policy. High industry interest.
2016	ļ	engrues.					engare policy. nigh industry interest.
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Table IX-1 (Continued)

PROGRAM A: EFFECTS OF H2 ON MATERIAL.

			Relates to:	3	-	Γ
Project Description	Activity Type	Fuel	Problem Area	Table	Item No.	R
Hydrogen gas connect Effects on General Engines.ing Materials, -00°+ to +160°F with extension to bigher temperatures for some applications.	Materials testing and interpretation.	II ₂	Fuel supply systems for industrial com- mercial, and residen-	1	1.1.1 5.1.1 6.1.1	М
iow-cycle fatigue, fracture toughness, crack		 	and E.C. engines.	7.1.1	7.1.1	
Materiris: low-and medium-strength steels, plastics, reinforced plastics, fiber-reinforced cements. Variables: temperature, H ₂ pressure, contaminants, exposure time, strain rate, stress level, stress intensity, material composition and		M ₂	equipment.	4	3.1.1 4.1.1 5.1.1 5.1.3 5.1.4	Lo
,		H ₂	Materials of existing pipelines.	3	1.1.1	In
		H ₂	Steels for new pipe- lines.	3	2.1.1	In
		Н ₂	Candidate nonmetallic pipeline materials.	3	2.1.4	In
		H ₂	Hydride containers.	3	4.8.1	Mo
		H ₂	Large and small pressure vessels (metal).	3 4	3.1.1 2.1.1 2.2.1	Мо
		li ₂	Large and small pres- sure vessels and tanks (reinforced plastics).	4	2.3.1	Мо
Hydrogen Permeability Studies: (a) Of high temperature alloys and permeation- resistant coatings up to 1500°F and 5000 psi.	Waterials research and development.	Н ₂	(a) Stirling cycle engines.	7	6.3.2 6.3.3	Мо
(b) Of coatings of Cd, Pb, Sn, glasses etc. at -60°F to -160°F and 1000 to 3000-ps1 H ₂ .		Н ₂	(b) Transportation and storage of H ₂ gas.	3 4	2.1.3	Lu
Role of Surfaces in Adsorption and Dissociation of Hydrogen. Study of the effects of surface oxides, sulphides and other comtaminant films on the mechanisms of hydrogen entry into metals.	Fundamental research.	Н ₂	Use of ${\rm H_2}$ in existing pipelines.	3	1.1.3	Lo
	Hydrogen Permeability Studies: (a) Of high temperature, welds. Hydrogen Permeability Studies: (a) Of high temperature alloys and permeation-resistant coatings up to 1500°F and 5000 psi. (b) Of coatings of Cd, Pb, Sn, glasses etc. at -60°F to -160°F and 1000 to 3000-psi H2. Role of Surfaces in Adsorption and Dissociation of Hydrogen. Study of the effects of surface ovides, sulphides and other comtaminant films on the mechanisms of	Hydrogen cm . onwent Eifects on General Enginesiig Materials testing Materialsor	Hydrogen part content Effects on General Enginessing Hadrogen part content enginess Hadrogen Hadrogen part content enginess Hadrogen Hadrogen	Activity Type Fuel Problem Area	Relates to: Section Activity Type Rydroken gav. comeant Effects on General Engines,,,	Activity Type Fuel Problem Area Activity Type Fuel Problem Area No. Item No. Bydrokan in

Table IX-1 (Continued)

PROGRAM A: EFFECTS OF H2 ON MATERIALS

n N		Relates to:	1	nce to			
		nerates to.	360010	M VIII			
ctivity			Table		Relevance	l .	
₹ Type	Fuel	Problem Area	No.	Item No.	to DoD	Priority	Remarks
rials testing interpretation.	Н ₂	Fuel supply systems for industrial com- mercial, and residen- tial equipment, I.C. and E.C. engines.	1	1.1.1 5.1.1 6.1.1 6.2.1 7.1.1	Moderate	4	Adequate materials can probably be selected for most case: on the basis of available knowledge.
数据的运动设计 计可可能分类 化氯化丁基丙烯基甲基丙烯基甲基丙烯基甲基甲基甲基甲基甲基甲基甲基甲基甲基甲基甲基甲基甲基	H ₂	Process and ancillary equipment.	2	2.5.2 3.1.1 4.1.1 5.1.1 5.1.3 5.1.4 3.5.2	Low	3	Materials for some process equipment may require testing at temperatures above 160°F or below -60°F. In most case: adequate materials can be selected on the bases of vailal knowledge.
	H ₂	Materials of existing	3	1.1.1	Indirect	1	Indirectly relevant to DoD, but very important to the
	H ₂	Steels for new pipe- lines.	3	2.1.1	Indirect	1	general introduction of H ₂ fuel system. Indirectly relevent to DoD, but very important to the general introduction of H ₂ fuel system.
	Н ₂	Candidate nonmetallic pipeline materials.	3	2.1.4	Indirect	4	Long-term pipeline development.
kaeskal	ч ₂	Hydride containers.	3	4.8.1	Moderate	4	Adequate materials can probably be selected on the basis of available knowledge.
	H ₂	Large and small pressure vessels (metal).	3 4	3.1.1 2.1.1 2.2.1	Moderate	2	Higher strength materials will need careful screening for use at high pressures. Needed to establish safety standards.
is of the subdividual field in	H ₂	Large and small pres- sure vessels and tanks (reinforced plastics).	4	2.3.1 4.1.1	Moderate	2, 4	Priority 2 for small high-pressure containers for DoD need (will be relevant to other gases). Priority 4 for large industrial vessels. Latter will probably be done by industry.
tals research velopment.	Н2	(a) Stirling cycle engines.	1	6.3.2 6.3 <u>.</u> 3	Moderate	Uncertain	Depends on DoD vehicle and small engine policy. High industry interest.
Karriston een	Н ₂	(b) Transportation and storage of H ₂ gas.	3 4	2.1.3	Low	4	Exploratory study only. If successful, study should be expanded and priority raised.
ental Feh.	H ₂	Use of H ₂ in existing pipelines,	3	1.1.3	Low	1	Previous studies of H ₂ entry into metals have used clean (or nominally clean) surfaces. In order to obtain results relevant to practical systems, it is necessary to obtain an understanding of the effects of surface films on H ₂ adsorption, dissociation and absorption. This work is particularly applicable to the use of H ₂ in existing pipelines but would obviously have wide general applicability and importance.
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PROGRAM A: EFFECTS OF Ha ON MATERIALS

				Relates to:	l .	ence to	Γ
Project	Project Description	Activity Type	Fuel	Problem Area	Table No.	Item No.	Rc
A-6	Vetal Hydride Etudies (a) Studies of H ₂ adsorption and desorption kinetics for candidate new alloying systems and modifications ·· existing systems. (b) Studies of cycle life of H ₂ storage beds as a function of: rates and depth of discharge; design, structure and condition of bed; contaminants in the H ₂ .	Materials research and development.	H ₂	Medium-scale and small-scale storage of ${\rm H_2}$.	4	7.1.3 7.1.4 7.1.5 7.1.6	Мо
A-7	Full-Scale Testing of Pipes with Hydrogen (a) Sections of new and used pipes (including girth welds), typical of existing pipelines, API 5LX Grades 42 to 70, pressurized with pure and contaminated H ₂ at temperatures and pressures (both steady and fluctuating) corresponding to the most	Full-scale mater- ials engineering testing and inter- pretation.	Н ₂	(a) Safety of pipe- line transpor- tation of H ₂ in existing pipelines.	3	1.1.2	In
	severe service or line-test conditions. Detailed examination and comparison of pipe and weld materials before and after H ₂ exposure. (b) Extension of test serves to candidate new high strength pipe materials.		Н ₂	(b) Specification of new pipelines for H ₂ use.	3	2.1.1	In
A-8	General Vaterials Support to Engineering Development and Component Testing Programs. Testing of equipment, components, required to operate in H ₂ environments under actual or simulated conditions corresponding to most severe service. Examination of exposed and failed components. Vaterials selection and consultation. Specialized materials development programs.	Materials testing and evaluation; special materials development.	H ₂	Fuel supply systems.	1	1.1 2 1.2.1 1.3.1 2.1.1 3.1.1 4.1.1 5.1.1 5.2.1 5.2.2 6.1.1 6.2.1 7.1.1 7.3.1 8.11.2	Va
			H ₂	Other H ₂ using equipment.	1	1.3.1 2.1.1 3.1.1	Vai
			Н ₂	Process equipment.	2	6.3.1 2.3.2 2.5.2 3.1.1 4.1.1 5.1.1 5.1.4	Va
			112	Transportation systems.	3	2,2,1 4,7,1 4.8,13	Va
			H ₂	Storage systems.	4	1.2.1 3.5.1 3.5.2 5.1.1 7.1.1	Va

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Table IX-1 (Concluded)

PROGRAM A: EFFECTS OF H, ON MATERIALS

		Relates to:		once to			
ivity ppe	Fuel	Problem Area	Table	Item No.	Relevance	Priority	Remarks
is research lopment.	H ₂	Medium-scale and small-scale storage of ${\rm H_2}$.	4	7.1.3 7.1.4 7.1.5 7.1.6	Moderate	2	Applicable to utility peak shaving requirements and the on-board storage of H ₂ for vehicles. Continuation and extension of existing programs. Level of effort should depend on outcome of technoeconomic and prototype studies (items 7.1.1 and 7.1.2 in Table VIII-4).
ale mater- ineering and inter-	H ₂	(a) Safety of pipe- line transpor- tation of H ₂ in existing pipelines.	3	1.1.2	Indirect	1	Full-scale pipe testing under practical operating conditions is needed in addition to materials testing (Project Λ -3) determine safety of existing pipelines for use with H_2 .
Son .	Н2	(b) Specification of new pipelines for H ₂ use.	3	2.1.1	Indirect	1	Required of new pipelines are to be qualified for H_2 use.
testing tuation; translaterials ont.	Н2	Fuel supp±y systems.	1	1.1.2 1.2.1 1.3.1 2.1.1 3.1.1 4.1.1 5.1.1 5.2.1 5.2.2 6.1.1 6.2.1 7.1.1	Variable	Variable	In all programs concerned with the development of engineering or manufacturing systems using, producing, transporting, or storing hydrogen, it is imperative that adequame maternals expertise is available to and used by the engineering staff in order to ensure that costly errors in maternals selection and use will not imperil the safety or reliability of the equipment.
	H ₂	Other H ₂ using equipment.	1	8.11.2 1.3.1 2.1.1 3.1.1	Variable	Variable	As above.
فارتخصارتك ويسترك والمسترك والمتاوان المتواطعة	H ₂	Process equipment.	2	6.3.1 2.3.2 2.5.2 3.1.1 4.1.1 5.1.1 5.1.4	Variable	Variable	As above.
SC ではたできなき、 ではないではない。 ではないできない。	H ₂	Transportation systems.	3	2.2.1 4.7.1 4.8.13	Variable	Variable	As above.
en e	H ₂	Storage systems.	4	1.2.1 3.5.1 3.5.2 5.1.1 7.1.1	Variable	Variable	As above.
of the states of		IX-27	1				

PROGRAM B: EFFECTS OF HYDROGEN CARRIERS AND PARTIALLY OXYGENATED COMP

					1	ence to on YIII	
Project	i 	Activity		Relates to:	Table		Re:
No.	Project Description	Туре	Fuel	Problem Area	No	Item No.	Į,
B-1	Stress Corrosion Cracking of Steels in Liquid Ammonia. (a) Establish the limits of the phenomenon with respect to steel composition and microstructure; stress level and type of stressing; contaminant type and concentration.	fasic research.	NH ₃	Fuel supply systems for NH ₃ -fueled equipment.	1	1.5 1	Unx
	(b) Long-term delayed fatigue and fatigue tests of presently used and cand date materials for NH ₃ pipelines, storage tanks and fuel supply systems as a function of contaminants; steel type composition and heat treatment; stress level; stress	Materials testing and interpreta- tion.	NH3	M ₃ tran-portation	3	1.2.1 1.2.3 2 3.1 4 2 1 4.2.2	Unc
	intensity; temperature; NH ₃ pressure.		NH ₃	Ni. storage.	2 4	5 1 5 3.6.2 3.6.3	Luc
B-2	Full-Scale Testing of Pipes with Liquid Ammonia (a) Testing of sections of new and used pipe including girth welds, API 5LX Grades 42 to 70, pressure sed with pure and contaminated NH ₃ at temperatures and pressures (both steady and fluctuating) corresponding to the most severe service or ine-	Full-scale waterials engi- neering testing and unterpreta- tion	<u></u>	(a) Safety of pipe- line transporta- tion of liquid NH ₃ in existing pipelines.	3	1.2.2	Lov
	test conditions. (b) Extension of test series to candidate new high strength pipe materials; and transportation or storage tank materials.		VII.3	(b) Specification of pipelines, transportation and storage tanks for liquid-NH, use.	. 4	2.3.1	_ov unz
B-3	Hydrazine Compatibility Studies. Development and testing of highly inert materials or coatings for storage of hydrazine for periods up to 10 years. Long-term compatibility of elastomeric materials for seals, expulsion bladders, diaphragms and hoses.	materials develop- ment and testing chemical studios, design.	, N ₂ H ₁	یسار term جسمتage من hydrazine	÷	5.3.1	to to
B-4	Carbon Monoxide/Metal Interactions at High Pressures. Tensile, notched tensile, and fatigue tests of new and used pipeline steels after long-term exposure to high-pressure pure and contaminated CO to establish if any deterioration of their mechanical properties occurs.	Materials testing and interpreta- tion	~, ,	Materials of existing pipelines	3		NO.1

Table IX-2

BFFECTS OF HYDROGEN CARRIERS AND PARTIALLY OXYGENATED COMPOUNDS ON MATERIALS

			D	ence to		T	
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<u> </u>				on VIII			į
У		Relates to:	Table	1 1	Retenance		
	Fuel	Problem Area	No	Item Vo.	to DoD	Priority	Remarks
Ē			T				
Tarch.	NH ₃	luel supply systems for NH ₃ -fueled equipment.	1	1.5.1	Uncertain	5	Relevance to DoD depends on decisions concerning energy depot concept and choice of fuel. Priority would be raised if NH ₃ selected for DoD use or as general civilian fuel.
testing Freta-	NH ₃	NH ₃ transportation.	3	1.2.1 1.2.3 2.3.1 4.2. 4.2.;	Uncertain	1	Existing unsolved problems of importance to DoT and industry. Will be important for any general use of NH ₃ as a fuel.
	NH ₃	NH ₃ storage,	2 4	5.1.5 3.6.2 3.6.3	Ui certain	4	Prior tv would be raised if higher strength steels were needed storage vessels.
engi-	хн ^з	(a) Safety of pipe- line transporta- tion of liquid NH ₃ in existing pipelines.	3	1 2 2	Low	3	Also needed to assure safety of NH ₃ pipelines for nonfuel uses, and set maximum operating conditions. Priority based on thus need.
開始的日本 (中に川・中央) (1245年) ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・ ・	NH3	(b) Specification of pipelines, transportation and storage tanks for liquid-VH ₃ use.	3	2.3.1 4.2.2 3.6.3	Low or uncertain	5	Choice of materials will depend on results from project B-1
develop- testing; studies.		Long term storage of hydrazine.	4	5.3.1	Moderate to high	Uncertain	Compatibility of materials with hydrazine is more limited by effects of materials on the decomposition of hydrazine than by the effects of h-drazine on the materials.
testing reta-	co	Materials of existing pipelines.	3	1.3.1	Lou	5	Priority will increase if general use of CO fuel is projected. Work should be extended to new pipeline materials if any deterioration of properties of existing pipeline steels is discovered.
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Table IX-3 PROGRAM C: EFFECTS OF OXYGEN (BY-PRODUCT) ON MATERIA

				Table IX-3	ı		
		PROG	MAM C:	EFFECTS OF OXYGEN (BY		CT) ON MAT	ſER
						ence to	
roject o	Project Description	Activity Type	ruel	Relates to Problem Area	Table	Item No.	
-1	Mechanical Properties of Metals in High-Pressure Oxygen. Studies of the effects of long-term exposure of metals to high-pressure pure and contaminated O ₂ on their mechanical behavior and surface condition. Mechanical tests should include crack-growth rates and high- and low-cycle fatigue.			Oxygen pipelines and storage vessels.	3	2.4.2	
-2	Ignition Hazards in Oxygen Pipelines. Determination of ignition hazards for pipeline steels in actual or simulated pipe configurations and containing surface cracks or flaws, when subjected to high stress rates, impact, or fatigue in the presence of high-pressure.pure or contaminated \mathbf{O}_2 .	Materials engi- neering, testing and evaluation.	02	Safety of oxygen pipelines and storage vessels.	3	2.4.1 8.2.1	
-3	Materials Support for Engineering Studies. Supporting studies of materials aspects of engineering and system design, hazard analysis, economic analysis.	Materials con- sultation.	02	Feasibility of largescale O_2 transportation.	1 3 4	1.4.2 2.4.3 2.5.1 8.2.3 8.3.1	

Table IX-3

PROGRAM C: EFFECTS OF OXYGEN (BY-PRODUCT) ON MATERIALS

rity	Fuel	Relates to Problem Area		ence to on VIII Item No.	Relevance to DoD	Priority	Remarks
as research.	02	Oxygen pipelines and storage vessels.	3	2.4.2	Moderate to low	3	Apparent lack of fundamental knowledge concerning possible effects of long-term exposure of metals to high-pressure oxygen. Relevant to existing as well as projected oxygen usage. Priority based on existing and future usage.
as engi- testing quation.	02	Safety of oxygen pipelines and storage vessels.	3	2.4.1 8.2.1	Moderate to low	3	Needed to confirm safety of O_2 transportation and storage systems. Relevant to existing as well as projected oxygen usage. Priority based on existing and future usage.
con-	02	Feasibility of large- scale O ₂ transporta- tion.	1 3 4	1.4.2 2.4.3 2.5.1 8.2.3 8.3.1	Moderate to low	S	Large-scale ${\rm O_2}$ pipeline and storage system might be an important aspect of the overall hydrogen economy. An early technical and economic judgement on this question is desirable. See Table IX-10, Program K, Project K-2.

Table IX-4

PROGRAM D: HIGH-TEMPERATURE MATERIALS STU

					ence to
	Activity		Relates to:	Table	1
Project Description	Туре	Fuel	Problem Area	No.	Item N
Reactions of Water Vapor with Metals and Ceramics at High Temperatures Fundamental kinetic studies of reactions of high H ₂ O-content environments with candidate gas turbine materials, including Si ₃ N ₄ , SiC, Cb alloys, and oxidation-resistant coatings.	Basic research.	H ₂	High-temperature com- ponents of gas tur- binos, scramjets, MHD systems.	1	1.2.3 1.4.3 2.1.1 4.1.1
Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate future gas turbine materials in actual or simulated H ₂ /air and H ₂ /O ₂ combustion gases. Ma erials to be tested should include Ni- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si ₃ N ₄ and SiC.	Materials testing and evaluation,	Н ₂	High-temperature com- ponents of gas tur- bines, scramjets, MHD systems.	1	1.2.4 1.4.3 2.1.1 4.1.1
Erosion-Corrosion of High-Temperature Materials in High-Velocity, High-Temperature Water Simulation and investigation of combined effects of erosion and oxidation/corrosion expected if high- velocity water is passed through cooling channels in gas turbine combustors, vanes or blades, or other H ₂ /air or H ₂ /O ₂ combustion systems.	Materials testing and evaluation.	Н ₂	Watercooled high-temperature gas turbine or $\rm H_2/O_2$ combustion components.	1	1.2.5 1.4.3 3.1.2
	High Temperatures Fundamental kinetic studies of reactions of high H ₂ O-content environments with candidate gas turbine materials, including Si ₃ N ₄ , SiC, Cb alloys, and oxidation-resistant coatings. Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate future gas turbine materials in actural or simulated H ₂ /air and H ₂ /O ₂ combustion gases. Ma erials to be tested should include Ni- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si ₃ N ₄ and SiC. Erosjon-Corrosion of High-Temperature Materials in High-Velocity, High-Temperature Water Simulation and investigation of combined effects of erosion and oxidation/corrosion expected if high-velocity water is passed through cooling channels in gas turbine combustors, vanes or blades, or	Reactions of Water Vapor with Metals and Ceramics at High Temperatures Fundamental kinetic studies of reactions of high H2O-content environments with candidate gas turbine materials, including Si ₃ N ₄ , SiC, Cb alloys, and oxidation-resistant coatings. Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate future gas turbine materials in actual or simulated H2/air and H2/O2 combustion gases. Ma erials to be tested should include N1- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si ₃ N ₄ and SiC. Erosion-Corrosion of High-Temperature Materials in High-Velocity, High-Temperature Water Simulation and investigation of combined effects of erosion and oxidation/corrosion expected if high-velocity water is passed through cooling channels in gas turbine combustors, vanes or blades, or	Project Description Reactions of Water Vapor with Metals and Ceramics at High Temperatures Fundamental kinetic studies of reactions of high H ₂ O-content environments with candidate gas turbine materials, including Si ₃ N ₄ , SiC, Cb alloys, and oxidation-resistant coatings. Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate future gas turbine materials in actual or simulated H ₂ /air and H ₂ /O ₂ combustion gases. Ma erials to be tested should include Ni- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si ₃ N ₄ and SiC. Erosion-Corrosion of High-Temperature Materials in High-Velocity, High-Temperature Water Simulation and investigation of combined effects of erosion and oxidation/corrosion expected if high-velocity water is passed through cooling channels in gas turbine combustors, vanes or blades, or	Reactions of Water Vapor with Metals and Ceramics at High Temperatures Fundamental kinetic studies of reactions of high. H2-0-content environments with candidate gas turbine materials, including Si ₃ N ₄ , SiC, Cb alloys, and oxidation-resistant coatings. Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate fluture gas turbine materials in acturl or simulated H2/air and H2/O ₂ combustion gases. Ma erials to be tested should include Ni- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si ₃ N ₄ and SiC. Erosion-Corrosion of High-Temperature Materials in High-Velocity, High-Temperature Water Simulation and investigation of combined effects of erosion and oxidation/corrosion expected if high-velocity water is passed through cooling channels in gas turbine combustors, vanes or blades, or	Activity Type Relates to: Table Type Relates to: Table Type Recations of Water Vapor with Metals and Ceramics at High Temperatures Fundamental kinetic studies of reactions of high: H_2-O-content environments with candidate gas turbine materials, including Si_3N_4, SiC, Cb alloys, and oxidation-resistant contings. Creep, Fatigue and Oxidation Resistance of High-Temperature Materials in Hydrogen Combustion Gases Determination of high temperature mechanical and environmental endurance of present and candidate future gas turbine materials in actural or simulated H_2/air and H_2/O_2 combustion gases. Ma erials to be tested should include Ni- and Co-base superalloys, Cb-alloys, dispersion-strengthened alloys, protective coatings, Si_3N_4 and SiC. Materials testing and evaluation. H_2 High-temperature components of gas turbine in the protective of the prot

Table IX-4 PR. AM D: HIGH-TEMPERATURE MATERIALS STUDIES

and the same		Reference to Section VIII								
<u>. </u>	Relates to:	Table		Relevance						
Fuel	Problem Area	No.	Item No.	to DoD	Priority	Remarks				
H.	High-temperature com- ponents of gas tur- bines, scramjets, MHD systems.	1	1.2.3 1.4.3 2.1.1 4.1.1	High	1	Basic information needed for use of K_2 as a gas turbine fuel. High temperature K_2 O reaction with Si_2N_4 and SiC likely to be important for present studies concerned with use of these materials in large and small turbines. Basic studies will guide further materials and coating development.				
Miles de construir com es sous establistes de construir d	High-temperature com- ponents of gas tur- binos, scramjets, MHL systems.	À	1.2,4 1.4.3 2,1.1 4.1.1	High	1	Information needed for use of H_2 as an aircraft /uel. Project entails expansion of existing gas turbine materials R, D, and T programs to include modified environment due to the change of fuel from hydrocarbons to H_2 .				
H 2	Watercooled high tem- perature gas turbine or H ₂ /O ₂ combustion components.	3	1.2.5 1.4.3 3.1.2	Low	4	Need for project depends mainly on demonstration of the engineering frasibility of water cooling for hot components of stationary or marins H ₂ /2ir or H ₂ /O ₂ gas turbines. Loughlife nozzles for H ₂ /O ₂ rocket engines may have related erosion-corrosion problems, but endurance required is very different.				
	IX-30									
475.00 V V V V V V V V V V V V V V V V V V										

Table 1X-5

PROGRAM E: MATERIALS FOR SERVICE AT CFYOGENIC TEMPF

Project	Project Description	.lctivity Type	Fuer	Relates to:		rence to on VIII Ites No.
NOTE	Projects A-1 and A-2 include the measurement of the mechanical properties of materials in H ₂ environments at temperatures down to -22°F. At this temperature, the effect of H ₂ environments on the mechanical behavior of materials is likely to be less important than the influence of temperature. The low-temperature data from Projects A-1 and A-2 will therefore contribute to Program E. Properties of Materials at Cryogenic Temperatures Determination of physical and mechanical proporties of metallic, polymeric and composite materials at cryogenic temperatures as required to support:	Physical and		Liquid H ₂ pipelines, transport and storage vessels, transfer and delivery systems.	2 3	4.1.1 4.1.3 2.2.1 4.6.2 4.7.1 3.3.2 3.3.3 3.4.1 3.5.1 6.1.1 6.1.2 6.1.3 6.1.4 6.2.1 6.3.1
E-2	Development of Low-Expansion, High-Strength, Light-Weight Composites for Liquid Hydrogen Fuel Tanks and Transfer Piping Development of polymeric composite materials and structures using selected controlled orientations of graphite or advanced organic fibers to provide low or zero expansion coefficients and low thermal conductivity in high-strength, stiff, light structures.	Materials development.		Aircraft and other lightweight liquid hydrogen tambs and piping.	ક	4.6.2
E-3	Development of Improved Cryogonic Insulation Naterials and Systems Naterials development support to engineering development of improved internal and external injuication for (a) Improved inverse indulation systems for bulk liquid & storage systems. (b) Insulation systems for directaft and space vehicle liquid hydrogen tanks able to function with hot- face temperatures of 180°F, 350°F and 650°F.	Materials davelopment,	H ₂ Liquid	(a) bulk or transportation storage of H ₂ liquid. (b) Aircraft and space vehicle liquid H ₂ fuel tanks.	3 4	4.1,1 3.5,1 6.2.1 6.3.1 6.4.1

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Table IX-5 PROGRAM S: MATERIALS FOR SERVICE AT CRYOGENIC TEMPERATURES

PRO	GRAN S:	Tablo iX-5		ajenic Temp	eratures		
Lty	Fuel	Relates to: Problem Area	į.	ence to on VIII	Relevance to DoD	Priority	Remurics
PRO DE LA COMPANSION DE		Liquid H, pipelimes, transport and storage vessels, transfer and delivery systems.	2 3	4.1.1 4.1.3 2.2.1 4.6.2 1.7.1 3.3.2 3.5.3 3.4.1 0.5.1 6.1.1 6.1.2 6.1.3 C.1.4 6.2.2 6.3.1 6.4.1	Moderate to high	\uz:nble	Work related to sircraft liquid H, fuel tank materials is of special concern to DoD and high priority (Items 4-6.1.1 through 6.4.1).
Military Chinasa Andrews Andre	Liquia	Aircraft wid other lightweight liquid ay irogew tanks and piping.	3	4.6.2	High	2	Likely to be relatively high-cost materials. Low- sypansion cryogenic tank materials reduce chill-down stresses in tanks and associated structures. See also Table IX-9, Project J-4(c).
	II II 2 LIQUIC	(a) Bulk or transportation rtorage of the liquid. (b) Aircraft and space vehicle liquid it, fuel tanks.	35 -4	4.1.1 3.5.1 6.2.1 9.3.1 6.4.1	Weinte lingh	3	Relavent to DoD storage of liquid Mg aircraft fuel. Insulation requirements are dependent on vehicle design details.

	·	Pi	ROGRAM F	: M	Table IX-C	CELLS ANI	···	YZE.
!						l .	ence to on VIII]
		Acti ity		, he	elates to:	Table		P.
% 5.	Project Description	Туре	Fuer	<u> </u>	Problem Area	No.	Item No.	ا _ ا
F-1	'mproved Noble Metal Electrode Catalysts. Noble metal electrocatalysts with high activity and stability al lower catalysts loadings for: (a) Alkaline and acid fuel cell cathodes.	Materials and engineering development.	H ₂		Fuel cell effi- giorny and cost.	1	8.1.3 8.2.1	н:
1	(b) Solid polyrer electroly.e electrolyzer electrodes.			(8)	SPE electro- lyzers.	2	2.2.2	H:
F-3	Nonnoble Metal Electrode Catalysts. Nonnoble metal catalysts with high activity and high chemical and physical stability at 150°C for: (a) Atkaline and acta fuel cell cathodes and/or anodys, and alkaline electrolyze electrodes.	Matorials and engineering development.	11.2	(a)	Fuel ceil offi- ciency and cost.	1	8.1.1 8.1.2 8.2.2 2.1.6	н
	(b) Molten carbonate fuel cell electrodes.				Molten carbonate fuel cell life.	1	8.3.1	L
 	(c) Solid polymer electrolyte electrolyzer electrodes.		 	(c)	SPE electrolyzers.	2	2.2.2	Hi
F-3	General Electrocatalyst Studies. (a) Fundamental studies of the mechanism of electrocatalysis.	Basic research and screening studies.	Various	(a)	Elec rocatalysis.	1	8.11.3	Hi
	(b) Measurement of single electrode characteristics of candidate electrocatalyst materials.			(9)	Electrocatalyst selection.	1	8.11.4	нз
F-4	Special Purpose Slect-ocatalysts. (a) Higher activity anode catalyst and selective cathode catalyst for direct methanol fuel cells. (b) Electrocatalysts that would permit the direct	Catalyst development.	Meth- anol	L	Direct methanol fuel colls. Electrolytic	1 2	8.5.1 8.5.4 8.5.1	Mo
	electrolytic production of methanol.				production of me'nanol.			_
	(d) Electrocatalysts for regenerative fuel cells that		Hydra- zing H ₂		Hydrazine fuel cells. Regenerative	1	8.7.1	Mo
	ure inconsitive to potential cycling. (e) Anode catalysts with ingroved long-term activity		271	(2)	H ₂ /O ₂ fuel cells.	1	8.10.1	No
	for direct assenia fuel cells.		NH ₃	(6)	Direct Ni, fuel calls.	1	0.3.1	Ľ
1 /-5	Electrode 'laterials and Structure Development. (a) Improved hydrophobic notymer bonding materials for alkaline fuel cells for service > 150°C.	Materials and engineering	H ₂		Alkaline fuel ceil olectrodes.	1	8.1.4	Hi
	(h) improved gas-diffusion electrode structures with controlled porosity for alkaline fuel cells.	development.		(b)	Aikalino fuol cell olectrodes.	1	8.1.5	No
	(c) high surface area electrodes for alkaline elec- tredes.			(e)	Alkaline electro- lyzers.	2	2.1.5	u _o
	(d) Coramics that can control wetting angle in molten carbonate fuel cells.			(d)	Molten carbonate fuel cells.	1	3.3.2	Lo
	(e) Materials that can control wetting angle in direct methanol fuel cells.		Neth- anol	(e)	Methanol fuel cells.	1	8.5.3	No
	(f) that function electrode structures of dual elec- trode structures for reponerative fuel cells.		H ₂	(1)	Regenerative fuel cells.	1	8.10.2 8.10.3	Мо
	(g) Electrods materials and conductors with improved corrusion resistance.		1	(g)	Acid and alkeling electrolyzers.	2	2.1.1 2.4.i	20
2-5					fn-32			

Table IX-6 PROGRAM V: MATERIALS FOR FUEL CELLS AND ELECTROLYZERS

Livity					nce to			
Siv ity		Re	elates to:	Taole		Relevance		
уре	Fuel		Problem Area	No.	Iron so.	to DoD	Priority	Remarks
ype Is and Fing ent.	H ₂	(a)	Fuel cell effi- ciency and cost.	1	3.1,3 8.2,1	High	2	Lower noble metal loadings needed to reduce cost and resource availability. DoD may be able to accept high loadings for critical applications.
		(b)	SPE electro-	2	2.2.2	High	l	As above.
E-		 	lyzers.			ļ		
is and Fing	Н ₂		Fuel cell effi- ciency and cost.	1	8.1.1 3.1.2 8.2.2 2.1.6	High	1	Efficient nonnoble metal electrode systems are needed to reduce cost.
SELECTION OF BRIDE		(b)	Molten carbonate fuel cell life.	1	8.3.1	Low	4	Most rendily applicable to fossil isel reformates.
5 4		(c)	SPE electrolyzers.	2	2.2.2	High	1	Nonnoble metal catalysts needed to reduce cost.
2								
Learch and	Various	(a)	Electrocatalysis.	1	8.11.3	ii±gh	2	Long-range basic studies necessary to provide sound fund mental basic for theoretical understanding and future advances in fuel cell and electrolyzer technology.
		(b)	Electrocatalyst selection.	1	8.11.4	High	1	Provide essential basis for empirical selection of candidate electrocatalysts.
studies.	Meth- anol	Ĺ	Direct methanol fuel cells. Electrolytic	2	8.5.1 8.5.4 8.5.1	Moderate	3	Methanol fuels cells would be of greater importance to DoD if methanol were adopted as a general vehicle fuel. Conceptual process. Exploratory study may be worthwhile
		_	production of mathanol.				 	
en e	Hydra- zine	(c)	Hydrazine fuel cells.	1	8.7.1	Moderate	2	Would improve efficiencies of hydrazine fuel cells.
	H ₂	(a)	Regenerative $\rm H_2/O_2$ fuel cells.	1	8.10.1	Moderate	2	Regenerative fuel cells offer possible electric power storage system.
	NH3	(e)	Direct NH ₃ fuel sells.	1	8.8.1	Low	4	Relevance to DoD and priority could increase if energy depot concept with NH, fuel was developed.
s and	Hg	(2)	Alkaline fuel cell electrodes.	1	8.1.4	High	1	Higher temperature operation would increase efficiency.
ent.		(b)	Alkaline fuel cell electrodes.	1	8.1.5	Moderate	2	Alternatives to polymer bonded structures in (a) above.
Z Z		(e)	Alkaline electro-	2	2.1,5	Voderate	3	Would reduce electrode overpotentials.
n de de		(d)	lyzers. Wolten carbonate	1	8.3 %	Low	4	Equivalent of hydrophobic electrode structures used in
	Meth-	(0)	fuel cells. Methanol fuel	1	8 5,3	oderate	3	alkaline fuel cells. Needed to control methanol cross-over problem.
	anol	Ľ	cells.				<u> </u>	
	H ₂	(f)	Regonerative	1	9.10,2	Moderate	2	Alternative to electrocatalysts that are insensitive to potential cycling, Project F-4(d).
		(g)	fuel cells. Acid and alkaline	2	8.10.3	Low	4	Longer life electrode structures.
8	<u> </u>		electrolyzers.		2.4.1	L	! 	
and ing ant.			1x-32					

PROGRAM F: MATERIALS FOR FUEL CELLS AND ELECTROLS

							nce to
Project		Activity			Relates to:	Table	
No	Project Description	Туре	Fue1	ļ 	Problem Ares	No.	Item No.
F -6	Improved Matrix (Diaphragm) Mater: -: 5						
	(a) Low-cost, compact, high-conductivity matrix	Materials and	H.	()	matrix materials	1	8.1.6
	materials for alkaline fuel cell3 and	engineering	1 7	1	ior higher team	2	2.1.2
	electrolyzers, able to operate above 150°C.	davelopment.			peratures in	-	7.14
			1) 1	athaline alec-		
			i	}	trolyzers.		
	(b) Improved ion-exchange membranes for alkaline		1	(b)	Matrix *ystems for	1	8.1.7
	and acid fuel cells.		1		fuel colls and		
					electrolyzers,		
	(c) Ceramic matrix materials with controlled		1	(c)	Molten carbonate	1	6.33
	porosity and improved resistance to thermal		- 1		fuel cells.		
	cycling yor molten carbonate fuel cells.			_			
	(d) Matrix materials impervious to methanol.		Meth-	(a)	Methanor fuel	1	8.5.2
		İ	anol		celis		
	(e) Matrix materials impervious to hydrazine.		Hydra.	(e)	Hydrazine fuel	1	3.7.2
	•	1	Line	i	cells.		
	(f) Thin, stable, high-conductivity mattrix materials		H ₂	(f)	Inorganic solid	1	8.4.1
	for incrganic electrolyte feel cells.				electrolyte fuel		•
					cells.		i
P-7	Too Conduction Described to all Collisions		;				
	Ion-Conducting Properties of Solids.	Basic research	١.,		; 1	•	8.4.2
	(a) Inorganic solids with ion-conducting properties	and materials	1 112	(a)	Inorganic solid	1 2	5.3.1
	at moderate temperatures.	development.		•	olectrolyte fuel cells and elec-	2	3.1
		davaropaant.	į		trolyzers.		
	(b) Low-cost, solid polymer electrolytes with			(2)	SPE electrolyzers	2	2.2.1
	higher temperature capability, high ionic		1	(0)	pri cicciony, ers	-	
	conductivity, good pechanical and chemical	1		l			
	stability.	· ·	ı		1		
		 					
F-8	Other Materials of Construction.				}		
	(a) Cell materials with matching coefficients of	Materials and	H ₂) (a)	High operating	1	8,4,1
	expansion, seals, electrical contact materials	engineering	-	•	temperature.	ž	2.3.2
	for inorganic solid electrolyte fuel cells and	development.		<u> </u>			
	electrolyzers.						
	(b) Higher temperature, lcw-cost frame materials for			(5)	Alkaline elec-	2	2.1.3
	alkaline electrolyzers, and low-rost mothods of		İ	í I	trolyzers.		
	fabrication.			1	į		
			i - 1	L			

Table IX-6 (Concluded)

DEOGRAM F: MATERIALS FOR FUEL CELLS AND ELECTROLYCERS

	*	-						
			Table IX-	,				
	POGRAN :	F:)	MATERIALS FOR FUEL		nce to	YZERS	T	
			Relates to:	Sectio	n VIII			
ity	Fuel	<u> </u>	Problem Area	Table	ltem .do.	Relavance to DoD	Priority	Remarks
aty state of the s	H ₂	(a)	Marrix materials for higher 'emperatures in alkaline electrolyzers.	1 2	8.1.6 2.1.2 2.14	High	1	A key item in increasing the efficiency of alkaline fuel cells and electrolyzers.
		(b)	Matrix systems for fuel cells and electrolyzers.	1	8.1.7	Moderate	3	An alternative to electrolyte-returning matrix materials (a) above.
		(c)	Molten carbonate	1	8.33	Low	4	Present materials have limited life.
E	anol		Methanol fuel	1	8.5.2	Moderate	3	Would control methanol cross-sver.
	Hydrx- Zine	(8)	Hydrazine fuel colls.	1	8.7.2	Moderate	3	Would control hydrazine cross-over.
		(1)	Inorganic solid electrolyte fuel cells.	1	8.4.1	Low	5	Lower temperature cells (See Project F-7(a)) would be preferable.
arch	Н2	(a)	Inorganic solid electrolyte fuel colls and electrolyzers.	1 2	8.4.2 2.3.1	Moderate	2	Breakthrough, analogous to discovery of 3-alumina, needed to reduce operating temperature of inorganic solid electrolyte fuel cells and electrolyzers.
		(b)	SPE electrolyzers	2	2.2.1	Moderate	1	Higher efficiency and reduced cost SPE electrolyzers.
	H ₂	(3)	High operating temperature.	1 2	8.4.i 2.5.2	Low	5	Lower temperature systems generally preferred. Project F-7(a) could increase the importance of inorganic solid electrolyte systems.
		(b)	Alkaline eler- trolyzers.	2	2.1.3	Moderate	2	Presently used polysulphones are limited to 150°C. Increases in electrolyzer temperatures (See Project F-2(a) and F-6(a)) would need corresponding improvements in frame materials.
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Table IX-7 PROGRAM G: MATERIALS FOR HIGH ENERGY DENSITY BATTER

		1	PROGRAM G:	MATERIALS FOR HIGH !	ENERGY DI	ENSITY BAT
				Relates to:	Secti	ence To on VIII
roject o.	Project Description	Activity Type	Fuel	Problem Area	Table No.	Item No.
-1	Shape Change in Zinc Electrodes. Investigation of dendrite growth and methods for controlling it.	Busic matorials research and engineering uevelopment.	Zn	All electrically rechargeable Zn batteries.	1	9.1.1 9.3.1 9.4.1
-2	Ion-Conducting Solids. (a) Optimization of 3-zlumina electrolyte composition for Na/S battery for improved conductivity, and	Materials research; ceramic	Na	(a) Electrolyte for Na/S battery.	1	٤.10.1
	chemical, thermal and mechanical stability. (b) Search for new ion-conducting solids with good conductivity at moderate temperature.	processing.	Various	(b) New battery systems.	1	9.10.2 9.11.1
	ic) Improved cation and anion exchange membranes.		Various	(c) New battery systems.	1	9.11.1
3	Battory Electrode Catalysts and Substrates. (a) Dual function, or two separate catalysts for O ₂ evolution and O ₂ reduction.	Catalyst develop- ment.	Zn	Charge/discharge efficiency of Zn/ air, Zn/O ₂ batteries.	1	9.1.2
	(b) More active Cl ₂ electrode substrate for Zr/Cl ₂ battery.	Materials development.	3n	Polarization of Cl_2 electrode.	1	9.4.2
-4	Materials Compatibility with Battery Environments. Materials support for the solution of battery design and construction problems according to specific battery requirements. Includes: problems of corrosion by Cl ₂ , molten salts, liquid metals, and sulfur; materials with matching expansion coefficients; materials for cases, seals, electrical feedthroughs, current collectors, electrical insulators, separacors, etc; thermal insulation; electrode and diaphragm materials.	Materials dovelopment and selection.	Various	Battery design and construction.	1	9.2.1 9.5.1 9.5.2 9.6.1 9.7.1 9.7.2 9.7.3 9.7.4 9.7.5 9.7.6 9.8.1 9.9.1

Table IX-7 PROGRAM G: MATERIALS FOR HIGH ENERGY DENSITY BATTERIES

orti							
etivity		D-1-4 4		ence To			
		Relates to:		on VIII	D-1		
etivity			Table		Relevance	.	
Туре	Fuel	Problem Area	No.	Item No.	to DoD	Priority	Remarks
materials th and ering ment.		All electrically rechargeable Zn batteries.	1	9.1.1 9.3.1 9.4.1	High	1	Future of Zn/air battery depends on solving shape-change problem in Zn electrode.
als hh; ceramic	Na	(a) Electrolyte for Na/S battery.	1	9.10.1	Moderate	2	Needed for improvement of efficiency and life of Na/S battery.
ling.	Various	(b) New battery systems.	1	9.10.2 9.11.1	High	2	Continuation and extension of work in progress. This project is related to Project F-7 and might be combined with it.
	Various	(c) Now battery systems.	1	9.11.1	High	2	Continuation and extension of work in progress. This project is related to Project F-6(b), and might be conbined with it.
ent. in; ceramic ing. t develop-	Zn	Charge/discharge efficiency of Zn/ air, Zn/O ₂ batteries.	1	9.1.2	High	2	Needed to improve battery efficienc; and reduce catalyst loadings.
ls ment.	Zn	Polarization of Cl ₂	1	9.4.2	Moderate	3	Needed to improve power density.
ent.	Various	electrode. Eattery design and construction.	1	9.2.1 9.5.1 9.5.2 9.6.1 9.7.1 9.7.2 9.7.3 9.7.4 9.7.5 9.7.6 9.8.1 9.9.1	Variable	Variable	Materials requirements are generally specific to particular battery and materials. R, D, and T should generally be undertaken as integral part of battery programs, rather than as independent, materials-oriented studies.
		IX-34					

PROGRAM H: CATALYSTS (FXCLUDING ELECTROCATALYS

Table IX-8

				Relates to:		ence to
Project	Project Description	Activity Type	Fuel	Problem Area	Table	Item No.
H-1	Long-Life Hydrazine Decomposition Catalyst Fundamental studies to elucidate mechanism of deteriora- tion of iridium catalysts and alumina substrate; development of improved, long-life, mixed-metal catalysts.	Catalyst R and D.	Hydra- zine	Hydrazine rocket engines.	1	3.2.1 3.2.2
H-2	Combusion Catalysts. (a) Fundamental studies of mechanism of the catalytic oxidation of hydrogen, e.g., by transition metal carbides.	Catalyst R and D.	Н ₂	(a) Catalytic combustion of h ₂ for sace and water heating.	1	7.2.1
	(b) Development of low-cost, long-life H ₂ oxidation catalysts that are resistant to poisoning by contaminants.			(b) Catalytic combus- tion of H ₂ for space and water heating.	1	7.2.1
	(c) Development of low-cost, long-life, high-activity catalyst for oxidation of NH ₃ to N ₂ + H ₂ O without formation of NO x		NH ₃	(c) Catalytic combus- tion of NH ₃ for space and vater heating.	1	7.4.1
and delight security of the contract of the co	(d) Development of improved catalyst for dissociation of NH ₃ .		NH ₃	(d) Dissociation of NH ₃ for improved combustion and/or reconversion to H ₂ .	1	7.3.1
	(e) Development of base-metal catalysts for oxidation of CO and methanol.		CO, methanol	(e) Catalytic combus-	1	7.5.1
H-3	Reversed Shift Reaction Catalys'. Development of catalyst for reversed shift reaction for production of CO (and methanol) by hydrogen reduction of nonfossil CO ₂ .	Catalyst development.	CO, methanol	Production of CO and methanol from non- fossil sources.	2	8.3.1

Table IX-8

PROGRAM H: CATALYSTS (EXCLUDING ELECTROCATALYSTS)

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		Relates to:		once to on VIII			
itivity Type	Fuel	Problem Area	Table No.	Item No.	Relevance to DoD	Priorit,	Remarks
at R and D.	Hydra- zine	Hydrazine rocket engines.	1	3.2.1 3.2.2	High	2	For small, long-life hydrazine monopropellant control and accessory rocket engines.
et R and D.	Н ₂	(a) Catalytic combus- tion of H ₂ for space and water heating.	1	7.2.1	Low to moderat <i>e</i>	3	Would provide fundamental basis for Project H-2(b).
		(b) Catalytic combustion of H ₂ for space and water heating.	1	7.2.1	Low to moderate	3	Would permit efficient, low-temperature combustion of H ₂ for space or water heating, in many cases without the need for venting combustion products.
	NH ₃	(c) Catalytic combus- tion of NH ₃ for space and water heating.	1	7.4.1	Low	5	Relevance and priority ratings would increase if energy depot concept was developed using NH ₃ as the fuel.
	NH ₃	(d) Dissociation of NH ₃ for improved combustion and/or reconversion to H ₂ .	1	7.3.1	Moderat <i>e</i>	4	For several applications of NH ₃ as a fuel, including direct combustion, vehicle engines, and indirect NH ₃ fuel cells. Work may also relate to improved NH ₃ synthesis catalysts.
	CO, methanol	(e) Catalytic combus- tion of CO or methanol for space and water heating.	1	7.6.1	Low	5	Relevance and priority ratings would increase if energy depot concept was developed using methanol as the fuel. Priority would increase if methanol was generally adopted as nonfossil liquid fuel.
et General		Production of CO and methanol from non-fossil sources.	2	8.3.1	Low	3,5	Needed for economic production process if CO (priority or methanol (priority 3) adopted as general fuels.

PROGRAM J: MISCELLANEOUS MATERIALS DEVELOPMENT AND FAB

Table IX-9

				Relates to:	Referen Section	
Project No.	Project Description	Activity Type	Fuel	Problem Area	Table No.	Item No.
J-1	Fabrication of Components for Advanced Gas Turbine and Rocket Engine Cooling Systems. (a) Materials engineering support for the design and development of advanced cooling systems for H ₂ /air and H ₂ /oxygen fueled gas turbines, combustion chambers, vanes, and blades.	Materials engi- neering.	н ₂	(a) Higher H ₂ combus- tion tempera- tures.	1	1.2.7
	(b) Materials engineering support for the design and development of cooled, long-life, rocket engine combustion chambers and nozzles.			(b) High-pressure, very high-tem- perature environ- ment.	1	3.1.2 1.4.3 4.1.1
J-2	Fiber-Reinforced Cements and Concretes. Further development of cements and concretes reinforced with glass fiber or chopped steel wire for large pressure vessels or storage tanks. Materials may also be polymer-impregnated.	Materials and engineering development.	Various	Large pressure vessels and storage tanks.	2	5.1.3 2.4.1 3.4.1
J-3	Vented-Lining Type Construction for Hydrogen Pressure Vessels and Pipelines. Vented-lining design used for high-pressure process vessels for hydrogen service could provide protection of pipeline and storage tank materials from H ₂ environments. This method might be economically feasible if low-cost manufacturing and assembly methods could be developed.	Materials and engineering development.	H ₂	Pipelines and storage tanks,	3 4	2.1.2
J-4	Fabrication of Fiber-Reinforced Composites. (a) Development of low-cost rapid, on-site methods for fabrication of large pressure vessels, pipe, and storage tanks.	Materials and engineering development.	Various	(a) Pipelines, pressure vessels, and storage tanks.	3	2.1.5 2.3.2 3.3.1
	(b) Rapid, low-ccst methods for mass production of small and medium size high-strength, light-weight pressure vessels for high-pressure gas storage, and insulated cryogenic storage of liquid H2, liquid NH3, etc.		H ₂ gas H ₂ liq. NH ₃ liq.	(b) Small containers and associated transfer and delivery piping.	3	4.1.1 4.7.1 4.1.1 5.2.1
	(c) Development and fabrication of special low- or zero-expansion composite materials and structures employing controlled orientations of graphite or advanced organic fiber reinforcements.		H ₂ liq.	(c) Aircraft and space vehicle tanks; transfer and delivery piping.	3	4.6.2
J-5	Friction and Wear in Gaseous and Liquid Hydrogen. Study of friction and wear behavior of metals and plastics in gaseous and liquid hydrogen. Development of improved materials and materials combination:	Matorials research and development.	Н2	H ₂ compressors and pumps.	2	4.1.2

Table IX-9 PROGRAM J: MISCELLANEOUS MATERIALS DEVELOPMENT AND FABRICATION

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	,	Relates to:	Section				
lty	Fuel	Problem Area	Table	Item No.	Relevance to DoD	Priority	Remarks
engi-	Н2	(a) Higher hy combustion temperatures.	1	1.2.7	High	1	Development of advanced cooling methods is likely to make the principal contribution to the use of higher combustion temperatures in gas turbines. This work would extend existing development programs.
		(b) High-pressure, very high-tem- perature environ- ment.	1	3.1.2 1.4.3 4.1.1	High	1	Present application is for long-life, re-usable rocket engines but results will also apply to $\rm H_2/O_2$ combustion systems for high-temperature industrial steam systems and MHD systems.
and At.	Various	Large pressure vessels and storage tanks.	2	5.1.3 2.4.1 3.4.1	Moderate	3	Materials system has wide general applicability.
To a second ty	42	Pipelines and storage tanks.	3 4	2.1.2	Low	4	Vented-lining design used for high-pressure chemical process vessels for hydrogen service might be employed more widely if low cost manufacturing and assembly methods could be developed.
and	Various	(a) Pipelines, pres- sure vessels, and storage tanks.	3	2.1.5 2.3.2 3.3.1	Low	4	Prinarily of concern to utility industries.
では、 では、 では、 では、 では、 では、 では、 では、	H ₂ gas H ₂ liq. NH ₃ liq.	(b) Small containers and associated transfer and delivery piping.	3 4	4.1.1 4.7.1 4.1.1 5.2.1	Moderate	3	Relevant to Dob with regard to on-board vehicle and marine storage of new fuels, and fuel supply storage. Also relevant to high-pressure or cryogenic storage of other gases.
	H ₂ liq.	(c) Aircraft and space vehicle tanks; transfer and delivery riping.	3	4.6.2	Hig:.	2	Possibly a superior approach to on-board aircraft storage of liquid H2. Priority could be down-graded if initial efforts are discouraging. Likely to be high-cost materials.
and	Н ₂	H ₂ compressors and pumps.	2	4.1.2	Moderate	4	Present knowledge of friction and wear behavior of materials in $\rm H_2$ environments is limited.
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Table IX-10

PROGRAM K: TECHNOECONOMIC AND ENGINEERING FEASIBILITY AN

			Relates to:		Reference t Section VII	
Project		Activity			Table	T
No.	Project Toacription	Туре	Fuel	Problem Area	No.	Iten
K-1	Naterials Supply Requirements for Blactrolysers. A study of the materials requirements and materials availability for competing advanced electrolyses technologies. Materials would include construction materials, specialty polymers and ceramics, noble and ronnoble electrocatalysts.	Technoeconomic evaluation.	Varions	Planning of research and development programs for the pro- duction of H2.	2	2.5
 K-2 	Engineering Feasibility and Recordics of Bulk Distribution of By-Product Oxygen. Materials support for an engineering system and economic study of the general distribution of oxygen produced as a by-product of hydrogen production. Study should include an analysis of potential hazards.	Engineering and economic evaluation.	By-pro- duct O ₂ .	Strategy decisions	3	2.4
x -3	Technical and Economic Evaluation of the Storage of Hydrogen as Metal Hearide Cost/benefit analysis of hydrine systems compared with Liquid and gaseous B ₃ for various storage capacities and applications, including peak shaving and vehicle uses. Materials requirements and a major cost element.	Engineering and economic evaluation.	ii ₂	Medium- and small- scale storage of ${ m H_2}$,	4	7.1

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Table IX-10

PROGRAM K: TECHNOECONOMIC AND ENGINEERING FEASIBILITY AND EVALUATION STUDIES

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	Relates to:		Section VIII		1	1		
Activity			Table	İ	Relevance			
Туро	Fuel	Problem Area	Ko.	Item No.	to DoD	Priority	Rémorts	
Activity Type Inceconomic Tuetion.	Various	Planning of research and development programs for the production of H ₂ .	2	2.5.1	High		Information gathered would also relate to materials requirements for large-scale use of fuel cells. Of high relevance to bob because of policy decisions relating to defens: materials requirements.	
Leering Conomic Contion.	9pro- (ct	Strategy decisions	3	2.4.3	Mode rate		if study results are favorable, a follow-up study of the technical and economic factors involved in the use aspects of hydrogen/axygen fuel system. Distribution study is given first consideration because it is considered the key element in the overall H2/O2 system. General availability of low-cost oxygen would offect many Dob interests.	
mation,	К2	Medium- and small- scale storage of \mathbb{N}_2 .	4	7.1.1	High	•	Of high relevance to Dob with regard to use of hydrid for on-board vehicle or marine storage of H ₂ . Study needs to be done in depth. Published comparish are not considered adequate.	
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